General remarks

A strange ambiguity seems to prevail when one discusses the event horizon of a black hole. On the one hand, it is a monster gobbling up things which will never return. On the other, it is a harmless region, as curvature and other physical variables are finite there for the freely falling observer. Is the event horizon eventful or a neutral spectator? The contradiction arises due to the need for reconciling two opposite points of view: that of the freely falling observer and the asymptotic observer. The suggestion that they are complementary and we cannot listen to both, thus, seems an attractive idea. When all the dust has settled down, what has been chalked up on the board? We are no wiser as even the two-dimensional models have turned out to be not solvable in closed form. Some new ideas and a lot of new techniques have come up in the process. The black-hole complementarity seems a very attractive idea. However, the theorists working in this area (both particle physicists and general relativists) have to get a lot more confidence in their mathematical techniques before a consensus emerges. Can an observable prediction emerge? One can always hope. Maybe in the area of cosmology, where too we have an event horizon, a prediction may emerge!

- 1. Schwarzschild, K, Berlin, 1916, 189.
- 2. Kruskal, M. D., Phys. Rev., 1960, 119, 1743
- 3. Misner, C. W., Thorne, K. S and Wheeler, J A., Gravitation, Freeman, 1972.
- 4. Hawking, S. W., Nature, 1974, 248, 30.
- 5 Vaidya, P. C., Curr. Sci., 1943, 12, 183.
- 6 Raychaudhuri, A. K., Phys. Rev., 1955, 98, 1123.
- 7. Tipler, F., Phys. Rev. Lett., 1980, 45, 949.
- 8 Hajicek, P. and Israel, W, Phys Lett., 1980, 80A, 9
- 9. Bardeen, J., Phys. Rev. Lett., 1981, 46, 382.
- 10 York, Jr J. W., Phys. Rev., 1983, D28, 2929.
- 11. Hawking, S. W., Commun. Math. Phys., 1982, 87, 395.
- 12. Banks, T, Peskin, M. E. and Susskind, L., Nucl Phys, 1984, B244, 125.
- 13 Mandal, G, Sengupta, A. and Wadia, S., *Mod. Phys. Lett*, 1991, A6, 1685.
- 14 Wrtten, E, Phys. Rev., 1991, D44, 314
- 15 Callan, C. G., Giddings, S. B., Harvey, J. A. and Strominger, A., Phys Rev., 1992, D45, R1005.
- 16. Banks, T., Dabholkar, A., Douglas, M. R. and O'Laughlin, M., Phys. Rev., 1992, D45, 3607; Russo, J. G., Susskind, L. and Thorlacius, L. B., Phys. Lett., 1992, B292, 13.
- 17. Giddings, S. B., UCSBTH preprint, 1993.
- 18. Susskind, L., Thorlacius, L. B. and Uglam, Phys. Rev., 1993, D48, 3743.
- 19. Thorne, K. S., Price, R. H. and MacDonald, D. A., Black Holes: The Membrane Paradigm, Yale University Press, New Haven, CT, 1986, and the references therein.
- 20. Page, D. N., Black Hole Information, University of Alberta, Report No. Alberta-Thy-23-93, 1993.

New routes for the synthesis of organometallic reagents

Mariappan Periasamy

In recent years, organometallic reagents have been utilized in numerous functional group transformations and C-C bond-forming reactions in organic synthesis. However, several of these synthetic methods require reagents which are not readily accessible to practising organic chemists. We have undertaken research efforts to synthesize some of these useful reagents in situ from readily available starting materials for applications in organic synthesis. For example, hydroboration of olefins can be readily achieved by the CH₃COOH/NaBH₄ or (CH₃COO)₂Hg/NaBH₄ combinations¹⁻³. These reagent combinations work as good as or better than the exotic reagent systems previously employed in certain selective hydroborations^{3, 4}.

It has been known for some time that the reaction of l_2 with NaBH₄ in diglyme gives B_2H_6 gas which is also relatively pure compared to the reagent generated using F_3B : OEt, in place of I, (ref. 5). It appears that this

method has not been widely utilized for the generation of B₂H₆ because these authors utilized vacuum line techniques for the isolation of the reagent in a series of liquid N₂ traps^{5,6}. We have found that B₂H₆ can be readily generated from the I₂-NaBH₄ combination using the apparatus recommended for the BF₄/NaBH₄ reagent system^{6,7}. Several amine-BH₄ complexes, including chiral

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amine boranes, have been prepared following this method for synthetic applications^{7,9}.

$$I_2 + 2 \text{NaBH}_4 \xrightarrow{\text{Diglyme}} B_2 H_6 \xrightarrow{\text{R}_1 N} R_3 N B H_3$$

Catecholborane and its derivatives have been readily prepared for synthetic applications using the B_2H_6 generated in this way $^{10-12}$.

Several iodoborane complexes have been prepared for applications in hydroborations, reductions, iodinations and also in cleavage of carbamates, esters and ethers 13-16.

Recently, we have reported that the readily accessible I₂/NaBH₄ reagent system is useful for several synthetic applications^{17, 18}.

McKennon et al. 19 reported that the I₂/NaBH₄ reagent system is excellent for the reduction of amino acids to amino alcohols.

The reactive species of the I₂/NaBH₄ combination in the THF solvent would be BH₃: THF (ref. 18). Hence, it is expected that this reagent system will become the method of choice for applications in which the BH₃ complexes are required.

We have also developed several useful transition metal reagents in situ for synthetic applications using readily accessible starting materials (Figure 1)²⁰⁻²⁵.

Also, methods have been developed for the preparation of several useful metal carbonyl complexes for synthetic utilizations (Figure 2)²⁶ 32.

It is anticipated that several of these methods would become the methods of choice for the synthesis of these materials. Very recently, we have reported convenient

$$CoCl_2 \xrightarrow{NaBH_4} [1] \xrightarrow{aq. NaOH} [Co(CO)_4]$$

$$CoX_{2} \xrightarrow{NaH/MgBr_{2}/CO/RC \equiv CR} \rightarrow (RC \equiv CR)Co_{2}(CO)_{6}$$

$$CoBr_{2} \xrightarrow{Na/Naphthalene} [Co(CO)_{4}]^{-}$$

FeCl₃
or
$$Na/Naphthalene$$
 THF/CO
 THF/CO
[Fe(CO)₄]²⁻

Figure 2.

methods for the synthesis of the following useful chiral amino alcohol and diol³³⁻³⁵.

It is hoped that the simple, convenient methods of preparation of these useful materials will be helpful in further research developments in these areas. We are actively pursuing research activities on these topics.

- 1. Marshall, J. A. and Johnson, W. S, J. Org. Chem, 1963, 28, 595-596.
- 2 Hach, V, Synthesis, 1974, 340-342.
- 3 Narayana, C. and Periasamy, M, *Tetruhedron Lett.*, 1985, 26, 1757–1760.
- 4 Narayana, C. and Penasamy, M., Tetrahedron Lett., 1985, 26, 6361-6364
- 5 Freeguard, G. F and Long, L. H., Chem. and Ind., 1965, 471-472.
- 6. Lane, C. F, Chem Rev., 1976, 76, 773-799.
- 7. Narayana, C. and Periasamy, M, J. Organomet Chem, 1987, 323, 145-147
- 8. Narayana, C. and Periasamy, M, J Chem Soc. Chem. Commun, 1987, 1857-1859.
- 9. Kanth, J. V. B. and Periasamy, M, J. Chem Soc Chem Commun, 1990, 1145-1147.
- 10. Suseela, Y., Prasad, A. S. B and Periasamy, M, J. Chem Soc. Chem Commun., 1990, 446-447
- 11. Suseela, Y. and Periasamy, M., J. Organomet Chem., 1993, 450, 47-52.
- 12. Penasamy, M., Prasad, A. S. B. and Suscela, Y., Tetrahedron, in press.
- 13. Reddy, Ch. K. and Periasamy, M., Tetrahedron Lett., 1989, 30, 5663-5664
- 14. Reddy, Ch. K. and Periasamy, M, Tetrahedron Lett, 1990, 31, 1919-1920
- 15. Kanth, J. V. B., Reddy, Ch. K. and Periasamy, M., Synth. Commun, 1994, 24, 313-318.
- 16. Kabalka, G. W., Narayana, C and Reddy, N K, Synth Commun, 1992, 22, 1793-1798, and the references cited therein.
- 17. Kanth, J V. B and Penasamy, M, J Org. Chem., 1991, 56, 5964-5965.
- 18. Prasad, A. S. B., Kanth, J. V. B and Periasamy, M, *Tetrahedron*, 1992, 48, 4623-4628.
- 19 McKennon, M. J., Meyers, A. I, Drauz, K and Schwarm, M, J Org Chem, 1993, 58, 3568-3571

- 20. Satyanarayana, N. and Periasamy, M., Tetrahedron Lett, 1986, 27, 6253-6256
- 21 Satyanarayana, N and Periasamy, M., J. Organomet. Chem, 1987, 319, 113-117.
- 22. Rao, S. A. and Periasamy, M., J. Chem. Soc. Chem. Commun., 1987, 495.
- 23 Rao, S. A and Periasamy, M., J. Organomet. Chem., 1988, 342, 15-20.
- 24 Rao, S. A and Periasamy, M., J. Organomet. Chem, 1988, 352, 125-131.
- 25 Rao, S. A. and Periasamy, M, Tetrahedron Lett, 1988, 29, 1583-
- 26 Satyanarayana, N. and Penasamy, M, Tetrahedron Lett, 1987, 28, 2633-2636.
- 27. Devasagayaraj, A, Rao, S. A. and Periasamy, M., J. Organomet. Chem, 1990, 403, 387-391.
- 28 Devasagayaraj, A and Periasamy, M., Tetrahedron Lett., 1989, 30, 595-596.
- 29. Periasamy, M., Reddy, M. R. and Devasagayaraj, A., Tetrahedron, 1994, 50, 6955-6964.
- 30 Devasagayaraj, A. and Periasamy, M., Trans. Met Chem., 1991, 16, 503-504
- 31. Devasagayaraj, A and Penasamy, M, Tetrahedron Lett, 1992, 33, 1227-1228.
- 32. Periasamy, M., Devasagayaraj, A. and Radhakrishnan, U., Organo-metallics, 1993, 12, 1424-1428
- 33. Kanth, J V. B. and Periasamy, M., Tetrahedron, 1993, 49, 5127-5132.
- 34. Periasamy, M, Kanth, J. V. B and Prasad, A. S. B., Tetrahedron, 1994, 50, 6411-6416
- 35. Periasamy, M., Prasad, A. S. B., Kanth, J. V. B. and Reddy, Ch. K., Tetrahedron: Asymmetry, in press.

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Bamboos – Some newer perspectives

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Bamboos are a part of the life and culture of southeast Asian people. They are a sought-after industrial raw material of considerable economic importance. All possible methods for their optimum utilization need to be tried. Bamboos have a peculiar behaviour of flowering and seeding at the end of very long vegetative growth phases, the length of which is considered to be species-specific. This makes their perennial propagation by seeds and their improvement by hybridizations practically impossible. Some inherent properties of bamboos can be exploited profitably to overcome these difficulties. In vitro induction of flowering has vast potential in bamboos. Bamboos need to be viewed in a broader perspective.

Bamboos are a most useful group of plants, and are members of the grass family (*Poaceae*)¹. Two important characters which make (majority of) them distinct from

other grasses are: (i) woody perennial habit and (ii) peculiar flowering and seeding behaviour². Most woody bamboos flower and seed at the end of very long periods of vegetative growth³ (Figure 1). In India, bamboos are the major source of raw material for pulp and paper industries⁴. Besides, they are also used for a variety of

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