

UPTAKE OF METAL IONS BY PHOTOTROPHIC BACTERIUM

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ABSTRACT

The phototrophic bacterium (PTB) *Rhodospirillum* sp. was found to extract a variety of metal ions. Nearly 80% of cadmium, 60% of nickel and 100% of lead were absorbed. Older cells were able to remove these metals faster.

INTRODUCTION

IN addition to their usefulness in metal extraction from ores¹⁻⁵, microbes have proved to be excellent tools for recovering accumulated metals from solutions⁶⁻⁸. These procedures may have overlapping functions. The use of microbial cells as biosorbents for heavy metals offers a potential alternative to the existing methods for decontamination or recovery or both of heavy metals from a variety of industrial process streams^{9,10}. Microbes are useful in extracting metals from seawater also¹¹. In the present work, we describe the removal of lead (Pb), nickel (Ni) and cadmium (Cd) ions from solutions by a phototrophic bacterium (PTB) *Rhodospirillum* sp. isolated from the shoot apex of water hyacinth¹². Preliminary results were reported earlier^{13,14}.

MATERIALS AND METHODS

Cultures of *Rhodospirillum* sp. were maintained on a medium described by Kobayashi *et al*¹⁵.

For the metal uptake studies, experiments were carried out in 30 ml screw-cap bottles completely filled to maintain anaerobicity. Metal ions (Pb:25-75 ppm; Cd:15-30 ppm; Ni:15-30 ppm and Hg 1-5 ppm) were added to the medium individually and sterilized. Ten-day-old cultures were inoculated at 1% level (O.D. 0.30), controls were maintained without addition of any metal ions. All the tubes were incubated at 25°C and illumination was provided from 3 × 100 W incandescent lamps (3000 lux at the vessel surface) for 8 hr a day with occasional shaking. After 10 days of incubation, the growth was measured at 665 nm in a Systronic 106 spectrophotometer. The cells were centrifuged, washed, dried, weighed and digested. The digested cells, culture filtrate and washing were analysed for metals by using an atomic absorption spectrophotometer¹⁶.

To test the ability of older cells in the uptake of metals, cells grown in control medium for 10 days were centrifuged, washed and reintroduced into

tubes containing the metals and further incubated for 3 days under conditions as mentioned above. After incubation, the cells were centrifuged, washed, dried, weighed, digested and tested for the metals. In each study, 10 tubes were kept for each metal at various levels as mentioned and the experiments were repeated 6 times.

RESULTS

The uptake of various metals is given in table 1. Hg even at 1 ppm level was toxic to the bacteria. Pb promoted the growth of bacteria as shown by increased biomass at 25, 50 and 75 ppm. Nearly 100% of the Pb added was taken up when added at 25 and 50 ppm levels initially. Only 42% of the added Pb was taken up by the cells at 75 ppm level.

Ni promoted the growth to an insignificant level and nearly 60% was taken by the cells. Cd did not enhance the growth of the bacteria but nearly 80% uptake of the added Cd was observed. Above 30 ppm levels Cd and Ni were lethal to the bacteria.

Table 1 Absorption of various metals by *Rhodospirillum* sp.

Metals added (ppm)	Dry biomass (mg/l of culture)	Levels of metal (ppm)			% absorption by cells	
		absorbed in filtrate washing cells				
Control	333.0	-	-	-		
Lead	25	625.0	N.D	N.D	24.8	100.0
	50	768.0	N.D	N.D	49.8	100.0
	75	750.0	46.0	N.D	28.0	42.0
Nickel	15	340.0	1.5	N.D	10.5	70.0
	25	495.0	3.3	3.0	18.7	74.0
	30	253.8	3.3	0.4	25.6	85.3
Cadmium	15	370.0	0.7	1.4	13.0	86.6
	25	330.0	N.D	3.7	21.3	85.2
	30	348.0	0.9	1.4	26.3	87.6

N.D: Not detectable.

Table 2 Absorption of metals by 10-day-old cells at the end of 72 hr after transfer from control culture

Metals added (ppm)	Dry biomass (mg l of culture)		Levels of metals in cells (ppm)	% absorption by cells
	Initial	Final		
Lead	50	336.4	350.4	45.0
	75	300.4	313.8	64.0
	100	360.6	381.2	72.0
Nickel	15	199.6	206.2	14.5
	30	287.3	296.4	25.1

Another significant observation (table 2) was that when a large biomass of older cells was introduced into the media containing these metal ions, the cell took up Pb and Ni at higher levels than when grown entirely on the medium containing respective ions. 80% of Ni and 90% Pb were taken up in 72 hr at 50 ppm level but the uptake declined at higher levels. The older cells were not able to take up Cd.

DISCUSSION

Though the utilization of PTB for H₂ production¹⁷ and in waste-water treatment¹⁸⁻²⁰ is well known, this is the first report on the capacity of PTB to uptake heavy metals. This study shows that by using solar energy, it is possible to clean up industrial water and recover metals, while a valuable fuel can also be produced.

Since there are many potential ligands including carboxylate, amine, phosphate, hydroxyl, sulphhydryl and other functional groups²¹⁻²⁵ on the cell membrane, most cells have a negative charge. These may be the possible sites of uptake of metals, which can attract positively-charged metal ions. These electrostatic forces and other undefined cell surface mechanisms for metal accumulation, are fast, reversible and useful, even if the cells are quiescent or dead. This may be one reason why the absorption of metal ions was more in older cells.

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