SYNTHESIS AND PURIFICATION OF OLIGODEOXYRIBONUCLEOTIDES: A MODIFIED APPROACH

KUMUD MAJUMDER, LATHA P. KADALAYIL and S. K. BRAHMACHARI*

Molecular Biophysics Unit, Indian Institute of Science, Bangalore 560 012, India.

ABSTRACT

A reduction in monomer consumption in the solid phase synthesis of oligodeoxyribonucleotides has been achieved by introducing a repeat coupling approach. Additionally, a method of anion exchange HPLC- purification of the synthesized nucleotides at room temperature has been developed in which a highly volatile buffer was used as the eluent. An alternative approach of HPLC purification of oligonucleotides is also reported.

INTRODUCTION

A typical solid phase synthesis of oligodeoxyribonucleotide, an important ingredient of genetic engineering, involves two essential steps, viz (i) chain building on the solid support through coupling reactions and (ii) HPLC purification of the synthetic product. Although, both these steps are well documented in literature¹⁻⁹ very little attention has been paid to economize these procedures. Here we present an economized approach to the synthesis and a convenient and easier method of HPLC purification of the synthesized sequences.

MATERIALS AND METHODS

Apparatus:

The Beckman 342 gradient HPLC system comprising of two 112 pumps, a 340 organizer, 420 system controller, 210 sample injector and 165 variable wavelength detector was used for all studies. Signals for two different wavelengths viz 260 and 280 nm for analytical and 260 and 295 nm for the preparative mode, were simultaneously recorded using a double channel strip chart recorder. The anion exchange (AE) HPLC purification was carried out using ultrasphere AX-10 or Partisil SAX columns (25 cm long, 4.6 mm i.d., 10 μ m particle size). A C₁₈-reversed phase (RP) ultrasphere ODS column (25 cm long, 4.6 mm i.d., and 5 μ m particle size), in conjunction with a precolumn has been used for the RP-HPLC analysis.

Materials:

Formamide was from Fluka AG. Common reagent grade acetonitrile, triethylamine (TEA), analytical reagent grade acetic acid and ammonium

acetate were from BDH, E.Merck and SM chemicals (all of Indian make). Both acetonitrile and TEA were distilled before use. The 2M stock solutions of TEAA (pH 6.8) and TEAB (pH 7.2) were prepared as published elsewhere¹⁰. Since TEAB buffer tends to change its pH on storage, CO₂ was bubbled just before its use.

Mononucleoside-phosphoramidites were prepared by suitable modifications of the existing methods. Oligonucleotides were synthesized on silica support using phosphoramidite approach. Nucleotide chains were removed from the silica support and subsequent deacylation was carried out by ammonia treatment. Oligomers were dissolved in the elution buffer or formamide for HPLC purification. Denaturation prior to injection of the oligonucleotide solution as such or in the presence of 80% formamide was effected by heating it to 90–95°C for 3–5 min.

Radioactive labelling and electrophoresis:

The oligonucleotides were end-labelled using $[\gamma^{-32}P]$ -ATP and analysed by polyacrylamide gel electrophoresis according to Maniatis et al¹¹.

RESULTS AND DISCUSSION

Synthesis of oligonucleotides by repeat coupling approach:

Solid phase synthesis of oligodeoxyribonucleotides using a manual open column demands that a large excess of the incoming monomer (10-20 times in excess over the support bound nucleotide chain^{1.5} bearing 5' - protecting group) be used to get a high coupling yield ($\geq 95\%$), of which only 5-10% is actually used in the coupling reaction. We have earlier developed a repeat coupling approach¹⁰ the usefulness of which is further demonstrated here through the synthesis of d(TIAA), d(TATA),

^{*}For correspondence.

Table 1	Synthesis	of	deoxyoligonucleotides	Ьy	repeat	coupling	approach	using		
phosphoramidite chemistry										

5'-3' Sequence	Specific Coupling step ^a	Ratio of incoming monomer & support bound chain ^b	Coupling yield in per cent ^c	Overall yield based on 1st base
d(TTAA)	A2 to A1	.5	89	
	T3 to A2	7	91	
	T4 to T3	$(5+1)^d = 6$	94	76
d(TATA)	T2 to A1	$(4+1)^d = 5$	95	
	A3 to T2	$(6+1)^d = 7$	96	
	T4 to A3	$(3+1)^d = 4$	94	86
d(GGCC)	C2 to C1	$(6+1)^d = 7$	95	
	G3 to C2	6	81	
	G4 to G3	6	85	65
d(CCGGCCGG)	G2 to G1	7	91	
	C3 to G2	7	88	
	C4 to C3	5	88	
	G5 to C4	10	94	
	G6 to G5	7	90	
	C7 to G6	$(5+1)^d = 6$	94	
	C8 to C7	$(5+1)^d = 6$ $(5+1)^d = 6$	95	53

Bases are numbered from 3' to 5' direction; Support bound chain refers to the sequence having 5'-OH groups after detritylation; Coupling yield for each step of chain elongation was determined on the basis of colour reaction of the dimethoxy trityl group at 495 nm³; Refers to repeat coupling.

d(GGCC) and d(CCGGCCGG) (table 1). The chain building for these oligomers was carried out manually (open column) on a silica support using phosphoramidite chemistry. It is clearly evident from table 1 that the repeat coupling approach enabled us to use only 4-7 fold excess of monomer with practically no concomitant loss of yield. This amounts to a 60-65% saving of the monomer per base addition.

Purification by anion exchange HPLC using volatile buffers:

The detrytilated DNA-octamer d(CCGGCCGG) was purified by the AE-HPLC method using TEAB or TEAA buffer. The sample injected was denatured either by heating or by adding a denaturant viz formamide. Figure 1 shows the chromatogram of the oligomer thus purified at 23°C (trace a) and 50°C (trace b). Comparison of these two profiles shows that the analysis can be efficiently carried out at either ambient or elevated temperature.

The efficacy of the AE-HPLC method was checked by injecting the once purified d(CCGGCCGG)onto the RP-C₁₈ HPLC column (figure 2) as the RP-C₁₈ method has been shown to be of comparable sensitivity to the two-dimensional method (homochromatography¹²) of determining sample homogeneity⁶. The observation of only one major peak in figure 2 shows that our AE-HPLC technique affords considerable degree of purification through a single injection of a fairly complicated mixture. The purified octamer, on autoradiography, showed a single spot (figure 3, lane b) which reaffirmed the purity of the compound. The sequence was further characterized by subjecting it to cleavage by the restriction endonuclease Hae III (GG+CC) (figure 4). As expected the digested product moved like a tetramer. The applicability of the present AE-HPLC approach to other self-complementary G-containing oligomers was tested by employing this technique for purifying the DNA tetramer, d(CCGG). Autoradiography of the once purified product (figure 3, lane a) showed the product to be homogeneous.

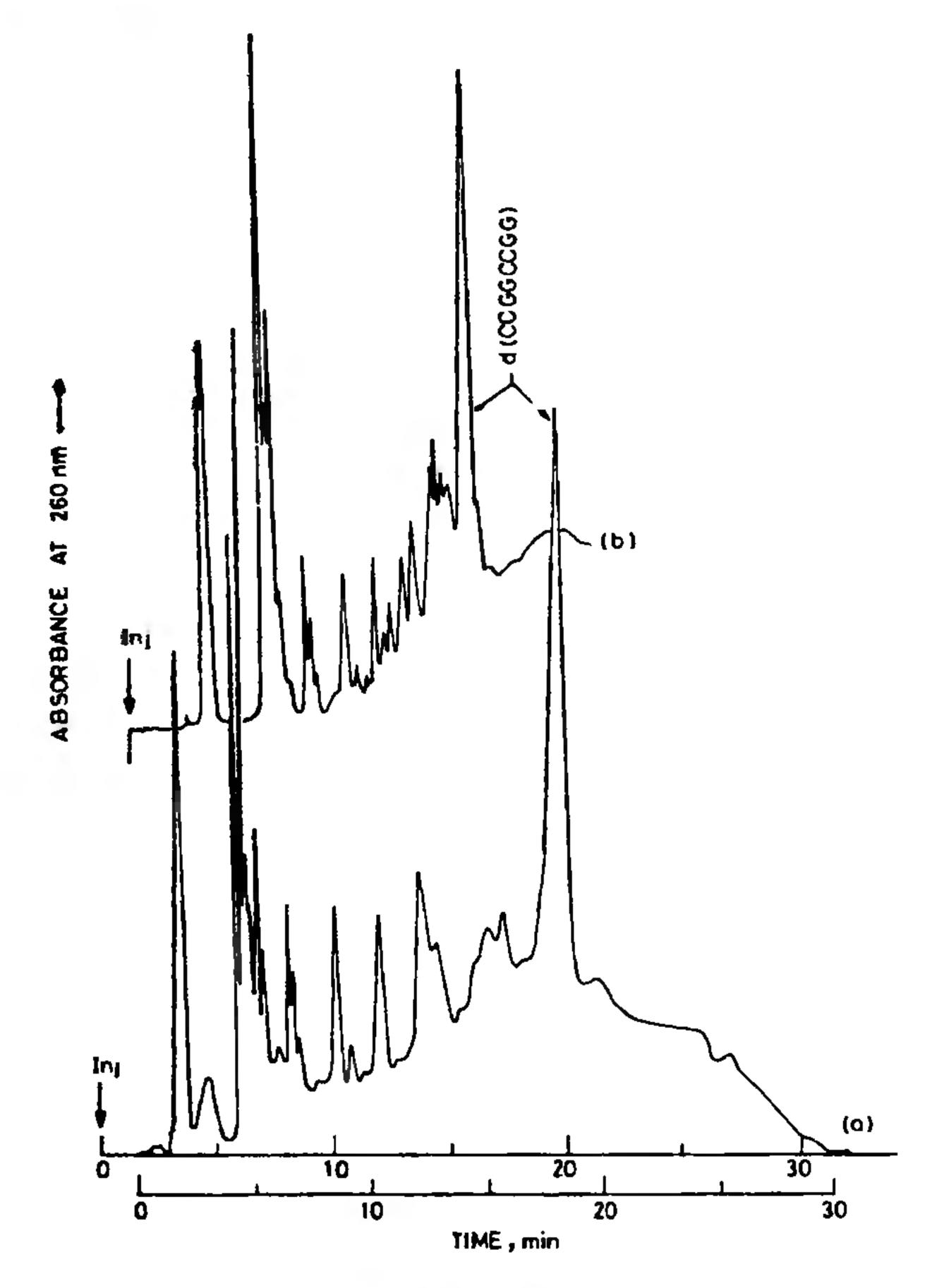


Figure 1. HPLC purification of the denatured d (CCGGCCGG). Detection: Absorbance at 260 nm. Range: 0.2 AUFS. Flow-rate: 1 ml/min. Column: Anion exchange-Ultrasil AX. For (a) — Eluent: 0-100% of 2M TEAB in water. Temp: 23°C (ambient). For (b) — Eluent: 0-100% of TEAA in water. Temp: 50°C.

Other common volatile buffers, like ammonium acetate, can easily substitute TEAA or TEAB of our method and this is illustrated in figure 5.

Anion-exchange columns are of importance to the purification of the G-rich self-complementary sequences as their recovery from reversed phase (RP) HPLC columns are often unsatisfactory⁶. Conventionally, the GC-rich sequences were purified by AE-HPLC at elevated temperatures (this being necessary to keep the oligonucleotide in the denatured state) using a non-volatile buffer^{7,8}. However, the usage of AE-HPLC columns at elevated temperatures with non-volatile buffers makes the recovery of the purified samples, especially of the smaller sequences, troublesome. Our method of injecting heat-denatured samples and purifying using a volatile buffer like TEAA or TEAB overcomes the limitations, otherwise commonly faced in AE-

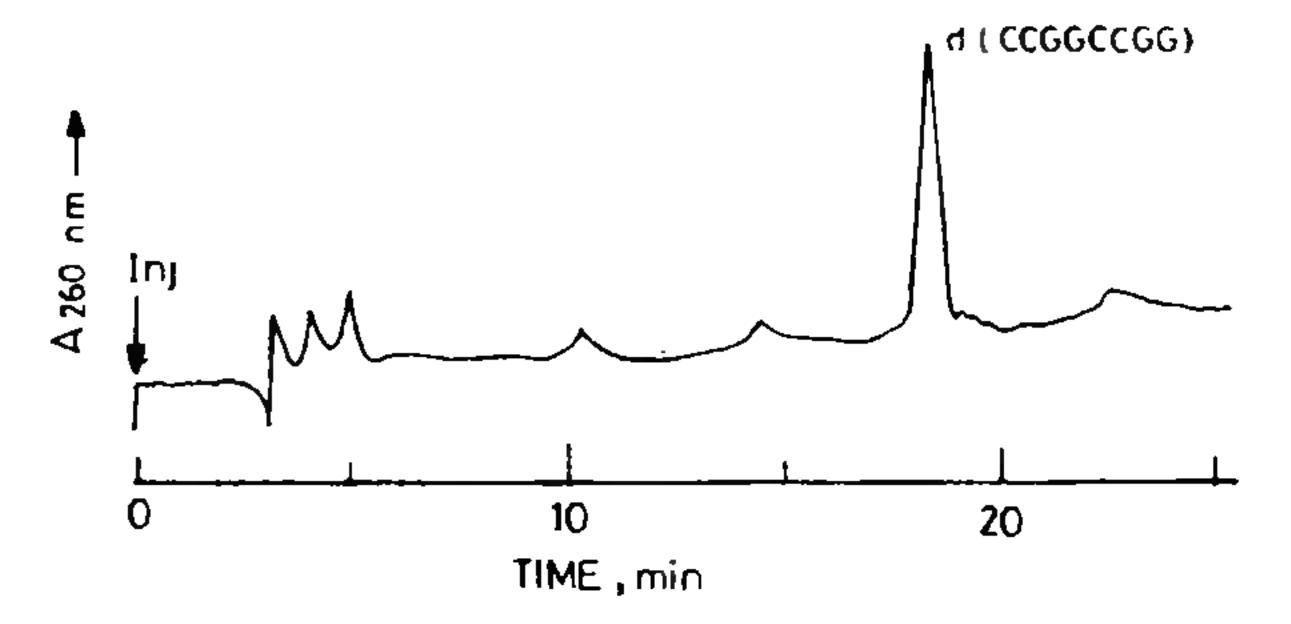
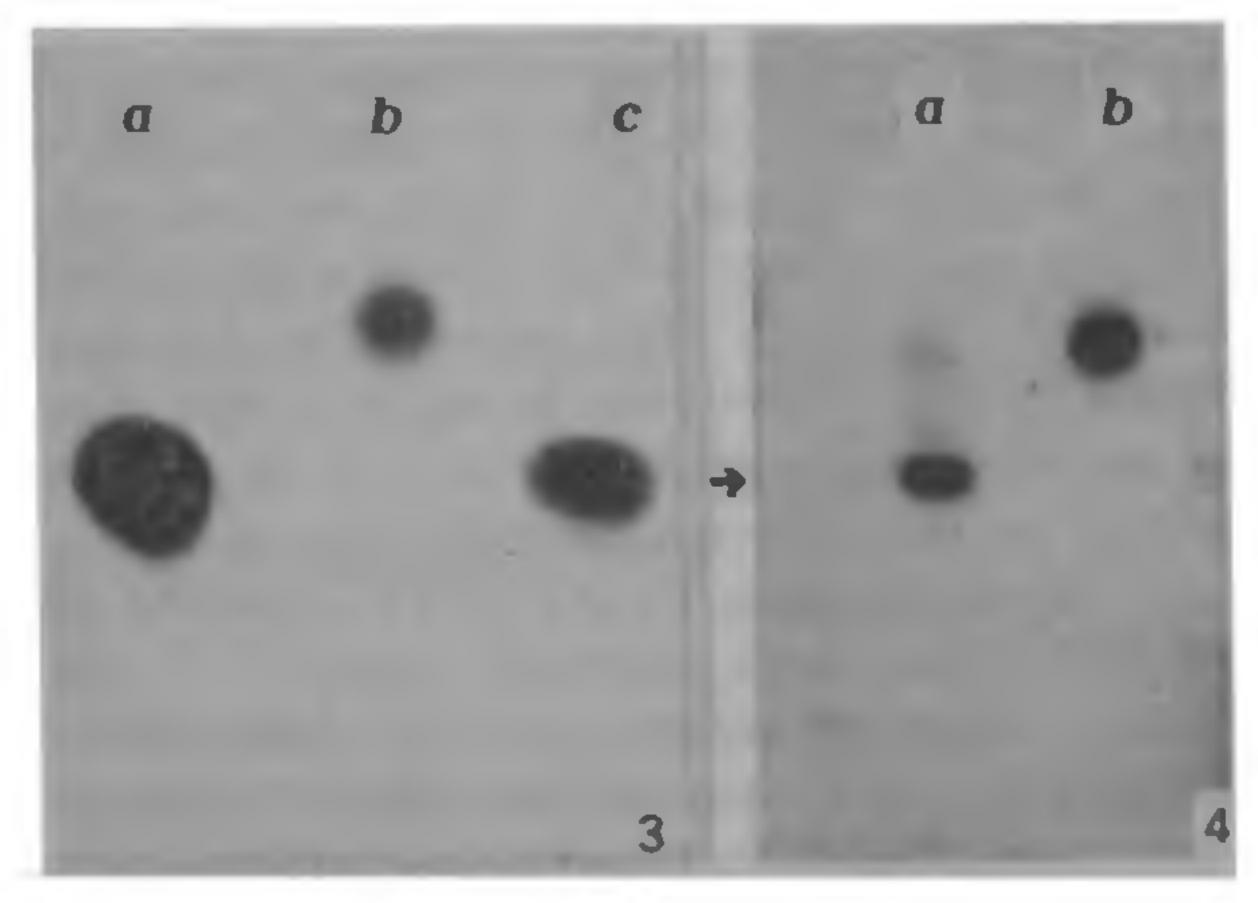


Figure 2. RP-HPLC analysis of AE-HPLC purified d(CCGGCCGG) Column: C_{18} -Reversed phase ultrasphere ODS. Eluent: Acetonitrile in 0.1M TEAA – 4.5% for 5 min, 4.5–8.5% in 16 min, 8.5–12% in 5 min. Temperature 50°C.

HPLC purifications using nonvolatile buffers. Since this method is equally effective at ambient temperature as well, the problems that might arise due to the use of high concentrations of buffers like TEAA at elevated temperatures are also overcome. We have found that use of high concentrations of TEAA at elevated temperature (≥50°C) can be detrimental to the silica-based columns. Further, we find that for dilute solutions of the oligomers where intermolecular interactions are minimal, addition of denaturant to the sample prior to injection is generally not necessary.

A similar method of using volatile buffer at ambient temperature for RP-HPLC has been reported by us earlier¹⁰.



Figures 3 and 4. 3. Autoradiogram of ³²P-labelled PAGE-urea analysed oligonucleotides. Lanes: (a) d(CCGG); (b) d(CCGGCCGG) and (c) d(GGCC); 4. Autoradiogram of Hae III digested ³²P-labelled d(CCGGCCGG) Lanes (a)—octamer digested with Hae III, (b)—octamer control. Arrow indicates the position of the tetramer marker.

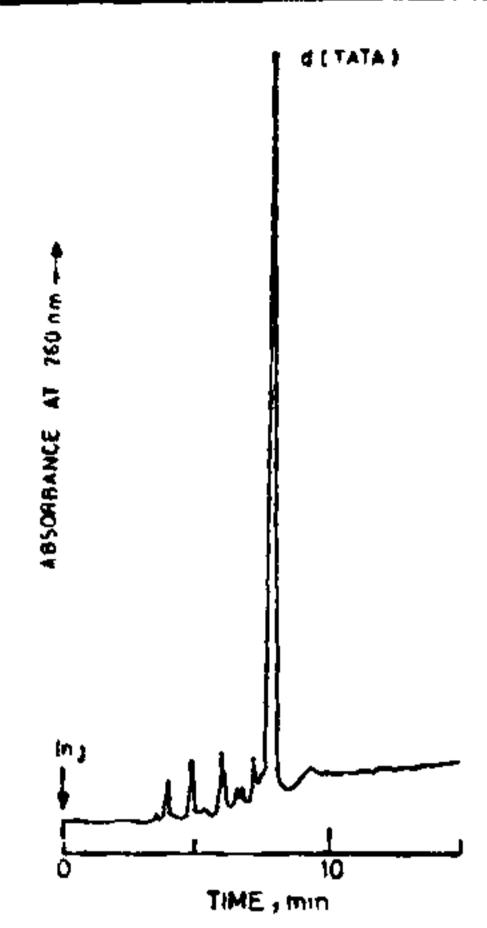


Figure 5. AE-HPLC analysis of d(TATA). Column: Partisil SAX. Eluent: A 0-100% gradient (in 15 min.) of 2M ammonium acetate (pH 6.5) in water. Temperature: 23°C (ambient).

An alternative HPLC approach:

The tetramer d(GGCC) was purified by using a combination of RP-HPLC and AE-HPLC. The nucleotide chains after removal from the silica support was injected on to an RP column and the elution carried out using a mixture of acetonitrile and TEAA. This initial purification takes advantage of the lipophilicity of the DMTr group which in a synthetic mixture is present only in the final sequence of the synthesis. The sequence of interest, therefore, becomes the longest retained species and hence easily identifiable. The once purified sample after detritylation with 80% acetic acid was subjected to ion-exchange HPLC following the method described earlier in this paper. The tetramer d(GGCC) thus purified was found to be very pure by autoradiography (figure 3, lane c). It may be pointed out that although the purification of detritylated oligonucleotides by AE-HPLC followed by RP-HPLC has been reported earlier^{8,1,5-16}, the reverse order of columns i.e. RP-followed by AE-HPLC, was considered to be difficult or even often impossible^{8,14–16}. This problem has been overcome in the present study by resorting to the RP-HPLC prior to the detritylation and then carrying out the AE-HPLC analysis.

CONCLUSION

In summary, a practical approach for the costefficient synthesis of oligodeoxyribonucleotides, especially for large scale batches of smaller sequences, has been developed. The method of injecting samples in the denatured state and their elution with volatile buffers not only overcomes the earlier mentioned limitations of the AE-HPLC method, but also makes this approach attractive for the purification of self-complementary oligomers. The alternative HPLC approach reported here was designed to exploit the known advantages of both the RP- and AE-HPLC methods.

ACKNOWLEDGEMENTS

The authors are grateful to Prof. H. Seliger, University of Ulm, FRG, for introducing them to the phosphoramidite chemistry. Thanks are also due to Rajani Srinivasan and K. Usha for the technical assistance received. LPK thanks CSIR, New Delhi, for a fellowship. Financial assistance from CSIR, New Delhi through a grant to SKB is gratefully acknowledged.

15 December 1986; Revised 13 March 1987

- 1. Matteucci, M. D. and Caruthers, M. H., J. Am. Chem. Soc., 1981, 103, 3185.
- 2. Fourrey, J. L. and Varenne, J., Tetrahedron Lett., 1984, 25, 4511.
- 3. Seliger, H., Klein, S., Narang, C. K., Seeman-Preising, B., Eiband, J. and Hauel, N., In: Chemical and enzymatic synthesis of gene fragments: A laboratory manual, (eds) H. G. Gassen, and A. Lang, Verlag Chemie GmbH, Weinheim, 1982, p. 81.
- 4. Rajendrakumar, G. V., Sundari, N. S. and Ganesh, K. N., Proc. Indian Acad. Sci (Chem. Sci.), 1985, 95, 357.
- 5. Dorman, M. A., Noble, S. A., McBride, L. J. and Caruthers, M. H., Tetrahedron, 1984, 40, 95.
- 6. Fritz, H. J., Belagaje, R., Brown, E. L., Fritz, R. H., Jones, R. A., Lees, R. G. and Khorana, H. G., Biochemistry, 1978, 17, 1257.
- 7. Gait, M. J., Matthes, H. W. D., Singh, M., Sproat, B. S. and Titmas, R., Cold Spring Harbor Symposia on quantitative biology, 1982, XLVII, 393.
- 8. McLaughlin, L. W. and Krusche, J. U., In: Chemical and enzymatic synthesis of gene fragments: A laboratory manual, (eds) H. G. Gassen and A. Lang, Verlag Chemie GmbH, Weinheim, 1982. p.177.

- 9. Gait, M. J. (ed.), Oligonucleotide synthesis: A practical approach. IRL Press, Oxford, 1984.
- 10. Majumder, K., Latha, P. K. and Brahmachari, S. K., J. Chromatogr., 1986, 355, 328.
- Maniatis, T., Fritsch, E. F. and Sambrook, J. In: Molecular cloning: A laboratory manual, Cold Spring Harbor Laboratory, CSH, 1982, p.122.
- 12. Tu, C. D., Jay, E., Bahl, C. P. and Wu, R., Anal. Biochem., 1976, 74, 73.
- 13. Newton, C.R., Greene, A.R., Heathcliffe, G.R., Atkinson, T.C., Holland, D., Markham, A.F. and Edge, M.D., Anal. Biochem., 1983, 129, 22.
- 14. Molko, D., Derbyshire, R., Guy, A., Roget, A., Teoule, R. and Bomcherle, A., J. Chromatogr., 1981, 206, 493.
- 15. Haupt, W. and Pingoud, A., J. Chromatogr., 1983, 260, 419.
- 16. Sonveaux, E., Bioorg. Chem., 1986, 14, 274.

NEWS

WHO ADVISORY COMMITTEE ON HEALTH RESEARCH

Since its inception in 1948, WHO has had close links with medical research and in 1959 the Advisory Committee on Medical Research (ACMR) was established to provide the Organization with scientific advice relating to its various research programmes. The work of this group of top-level international scientists, whose members serve for four years and have included many Nobel Prize winners, has thus continued for over a quarter of a century.

In 1986 the group was renamed the Advisory Committee on Health Research (ACHR), the change in name reflecting added emphasis on health protection and promotion without, however, neglecting the importance of strictly medical research. Health systems research, for example, gives practical support to the strategy of Health for All by the year 2000 and enhances the value to society of prevention, planning and optimal use of available resources.

The Regional Advisory Committees on Health

Research work through each of the six WHO Regional Offices. The global ACHR has the complex task of coordinating their activities. Despite inevitable differences between regions and between countries, the common aim is to focus on research where it will result in the most rapid advance towards the goal of Health for All.

The newly named ACHR met in Geneva in October 1986. One of the matters discussed was the transfer of technology among countries whose needs for health technology, the available resources, and the sociocultural situations differ widely; the Committee therefore strongly recommended that each country should evaluate new health technologies before they are introduced. The evaluation must take into account not only technical efficiency but also such matters as cost, relative priority, and acceptability by the population. (World Health Forum, 1987, Vol. 8, No. 1, p. 117; World Health Organization, 1211 Geneva, Switzerland).

MICROBES TO TACKLE PLANT DISEASE

Pests have a remarkable ability to become resistant to the chemicals used to control them. Developing new formulations is costly, and there are many questions about their effects on the environment and on food. In this context biological control of disease is gaining favour ecologically and economically. Future crop systems may well benefit

from research at Bristol University using one microorganism to fight another already in the soil and introducing mixed species of plants. (Spectrum, No. 205/ 1987/2; Spectrum British Science News; British Information Service, British High Commission, New Delhi 110 021).