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Product No.	$\mathbf{R_i}$	R ₂	R ₃	R ₄	R ₅	R ₆	% yield with KOH/DMF.	Observed m.p.	Literature m.p.
Illa	H	Н	H	Н	H	Н	85	52°	52°8
IIIb	H	H	Н	Н	OCH_3	Н	84	78°	78° ⁸
HIc	OH	H	Н	Н	OCH ₃	Н	84	95°	95°14
IIId	ОН	Н	Me	Н	OCH_3	Н	80	99°	98°13
IIIe	ОН	Н	Н	Me	Н	Н	78	99°	99°13
HIIf	OH	Н	Н	Me	OCH_3	Н	81	97°	98°13
IIIg	OH	Н	Н	Me	H	NO_2	60	187°	187° ⁸
IIIh	OН	Н	Н	Me	NO,	Н	67	204 °	204° ⁸
IIIi	ОН	NO_2	Н	Me	Η	Н	72	158°	158° ⁸
IIIj	OH	NO_2	Н	Me	OCH_3	H	65	210°	210°8
IIIk	OH	Вг	H	Me	H	H	78	108°	108°13
HIII	OH	Br	Н	Me	OCH_3	Н	82	148°	149°13
IIIm	OH	H	Н	Н	H	Н	80	89-90°	89–90°15

Satisfactory elemental analysis was found for all chalcones.

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A NEW ROUTE TO THE SYNTHESIS OF SOME SUBSTITUTED 3,4-BENZFLUORENES

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THE AlCl₃-catalyzed condensation of aromatic hydrocarbons with suitably substituted alicyclic lactones having a cyclohexane or a cyclopentane moiety, followed by cyclization, reduction and aromatization of the resulting alkylates, has a limited application in the synthesis of polycyclic compounds, due to extensive rearrangement of the resulting carbocation and polyalkylation¹. In this laboratory, we have developed a method for the synthesis of some substituted 3,4-benzfluorenes by utilizing the catalyzed alkylation of aromatic hydrocarbons with a substituted indanol derivative which will generate a stable benzylic carbocation in the presence of Lewis acid and thereby promote normal alkylation.

The indanol derivative investigated by us for the alkylation study has been 5-methyl-1-hydroxy-trans-2-indanylacetic acid (I) which has been prepared from 5-methylindan-1-one-2-acetic acid, m.p. 125°, semi-carbazone, m.p. 236°, methyl ester, m.p. 53°, synthesized from β -p-toluoylpropionic acid by an adaptation of the method of Roy². Sodium borohydride reduction of the keto-acid in alkaline solution according to House et al³ furnished the trans-acid (I) (80%), m.p. 138°, P.M.R.: δ , 4.21 (benzylic H, d, J = 8.8 Hz,

trans) along with the cis-lactone (20%), b.p. $160^{\circ}/2.5$ mm., m.p. 53° .

The AlCl₃-catalyzed alkylation of benzene with the hydroxyacid (I) in a mixture of sym-TCE and nitrobenzene at low temperature furnished the normal product, 5-methyl-1-phenyl-trans-2alkylation indanylacetic acid (II, R = H) in 65% yield, m.p. 115°, P.M.R.: dibenzyl proton at δ , 4.2, d, J = 9 Hz (trans), purified and isolated through the methyl ester, b.p. 190-195°/2 mm. Intramolecular Friedel-Crafts acylation of the acid (II, R=H), furnished the benzfluorenone derivative (III, R=H) in 66% yield, b.p. 160-165°/2 mm., which showed the characteristic trans dibenzylic-H at 4.2 δ , d, (J = 8.8 Hz) and the Ar-H, peri to carbonyl group, at $7.8 \,\delta$ in the PMR spectra. Reduction of the ketone by LiAlH₄ followed by dehydration and dehydrogenation furnished 7methyl-3,4-benzfluorene (IV, R=H), m.p. 70°, picrate, m.p. $136-137^{\circ 4}$, TNB-complex, m.p. $146-147^{\circ}$.

The catalyzed condensation of the hydroxyacid (I) with toluene in a similar manner afforded 5-methyl-1-p-tolyl-trans-2-indanylacetic acid (II, R=CH₃), m.p. 116°, methyl ester, m.p. 53°, in 60% yield. The structure of the acid was confirmed by an unambiguous synthesis starting from p-tolyl magnesium bromide and methyl 5-methyl-1-keto-2-indanylacetate followed by dehydration and catalytic hydrogenation, as well by the PMR spectra. Intramolecular Friedel-Crafts cyclization of the acid gave the benzfluorenone derivative (III, R=CH₃), m.p. 99-100°, 2,4-DNP derivative, m.p. 222°, which was converted to 7,1'-dimethyl-3,4-benzfluorene (IV, R=CH₃), m.p. 107°, picrate, m.p.

143°, TNB-complex, m.p. 165°. The PMR spectrum of the hydrocarbon shows two ortho coupled meso aromatic protons at δ 8.62 (d, J = 8.8 Hz) and at δ 8.24 (d, J = 8.8 Hz), far downfield from the rest.

The reaction is being fully investigated for the synthesis of highly substituted benzfluorenes and naphthofluorenes, some of which are of interest to us for a study of carcinogenic activity and from the standpoint of stereo-chemistry as they are likely to develop chirality due to molecular overcrowding.

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BIOMASS ENERGETIC YIELD AND MAINTENANCE COEFFICIENTS OF ASPERGILLUS WENTH ON DIFFERENT CARBOHYDRATES

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ERICKSON et al¹ applied mass and energy balance regularities to aerobic microbial growth processes by using equivalents of available electrons in growth substrate, biomass and product. This approach has been used in the present investigation for evaluating the efficiency of conversion of some carbohydrates to fungal biomass using Aspergillus wentii Wehmer strain Pt 2804. This fungus was selected as it fulfilled most of the criteria outlined by Imrie and Righelato² for selection of suitable strains for production of fungal biomass from carbohydrates.

The fungus was cultivated in a 5-litre bioreactor containing 3.5 litre of modified Mandels and Weber's medium³. The cultivation conditions in the bioreactor runs and the determination of growth associated