was found that 2-HANO is also a sensitive reagent for copper(II).

	Molar
	absorptivity
	1. mol ⁻¹ cm ⁻
Reagent	$(\times 10^3)$
Orthohydroxy acetophenone oxime ⁸	3.4
n-Pentyl-2-pyridylketoxime ¹	2.7
Phenyl 2-(6-methylpyridyl)	10.8
ketoneoxime ¹	
6-Methyl-2-pyridinecarboxamide- oxime ¹	7.2
Salicylaldoxime (n-amyl acetate)1	7.2
2-Hydroxy-1-acetonaphthone-	5.1
oxime 2,4-Dihydroxyvalerophenone-	0.16
oxime	

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- 1. Singh, R. B., Garg, B. S. and Singh, R. P., Talanta, 1979, 26, 425.
- 2. Marczenko, Z., Spectrophotometric determination of elements, Ist edn, John Wiley, Inc., New York, 1976, p. 247.
- 3. De, A. K., Khopkar, S. M. and Chalmers, R. A., Solvent extraction of metals, Ist edn, von Nostrand, London, 1970, p. 100.
- 4. Hussain, R. C. and Raju, N. A., Proc. Indian. Acad. Sci., 1976, A84, 144.
- 5. Hussain, R. C., Ph.D. Thesis, S.V. University, Tirupathi, 1977.
- 6. Vogel, A. I., A textbook of Quantitative inorganic analysis, 3rd edn, (ELBS & Longmans, London) 1961, p. 358.
- 7. Hussain, R. C. and Raju, N. A. Proc. Indian. Acad. Sci., 1976, A84, 144.
- 8. Reddy, T. S. and Rao, S. B., Curr. Sci., 1980, 48, 440.

SYNTHESIS AND PHYSIOLOGICAL EVALUATION OF A THIAZOLIDINEDIONE AND ITS AZODERIVATIVES FROM NINHYDRIN

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NINHYDRIN has been used to synthesise a variety of heterocyclics¹⁻³ and to estimate amino acids. Quinoxaline derivatives are known to possess insecticidal and acaricidal activity⁴. Several thiazolidine-diones are reported to exhibit a good spectrum of biological activity⁵⁻⁸. These reports prompted us to undertake the synthesis of a 2,4-thiazolidinedione 2-azine (III) containing 11H-indeno[1, 2-b] quinoxalin-11-one moiety and its aryl azoderivatives (IV) starting from II.

The starting materials 11H-indeno[1, 2-b] quinoxalin-11-one (I) and 11H-indeno[1, 2-b] quinoxalin-11-one thiosemicarbazone were prepared according to our earlier procedure⁹.

11H-indeno[1,2-b]quinoxalin-11-one thiosemicarbazone (II) on interaction with either chloroacetyl chloride or chloroacetic acid in the presence of anyhydrous ethanol and dry pyridine gave the cycloproduct 2,4-thiazolidine dione 2-azine (III) containing 11Hindeno[1,2-b]quinoxalin-11-one which on further reaction with diazolised substituted sulphanilamides furnished the corresponding p-[(2,4-dioxo-5-thiazolidinyl)azo]benzene sulphonamide 2-azine containing 11H-indeno[1,2-b]quinoxalin-11-one (IV). The products (III & IV) obtained have been confirmed by analytical and spectral (IR, PMR and mass) data.

2,4-Thiazolidinedione 2-azine containing 11H-indeno [1,2-b]-quinoxalin-11-one (III):

A mixture containing II (0.01 mol) and chloroacetyl chloride or chloroacetic acid (0.01 mol) in anhydrous ethanol (10 ml) and dry pyridine (10 ml) was refluxed for 6 hr and poured into crushed ice with constant stirring. The product separated was filtered, dried and purified from DMF, m.p. 250°C, yield 70% (Found: C, 61.88; H, 2.98; N, 20.10; $C_{18}H_{11}N_5OS$ requires C, 62.62; H, 3.18; N, 20.29% IR (nujol, vcm⁻¹) 3400 (-NH-), 1710 (amide C = O), 1600 (vCN); PMR (δ , pyridine-d₅), 7.5-8 (8H, m, aromatic), 2.20 (2H, s, -s- CH_2 -), 8.55 (s, 1H, -NH-); mass: m/z 345 (M⁺,

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 5°_{0}), 250 (52°₀), 217 (100°₀), 190 (34°₀) and 104 (20°₀).

p-[(2,4-Dioxo-5-thiazolidinyl)azo]henzene sulphonamide 2-azine containing 11H-indeno[1,2-b]quinoxalin-11-one (IVak

Sulphanilamide (0.025 mol) dissolved in aqueous acetic acid (8°₀. 25 ml) was diazotised with NaNO₂ (0.25 g in 30 ml) in an ice bath. The resulting diazosolution was added gradually with stirring to a previously cooled thiazolidine dione (0.025 mol) in aqueous NaOH (2°₀, 25 ml), kept in an ice bath. After the addition, the mixture was stirred mechanically for 1 hr during which the azo compound separated out. It was filtered and crystallised from acetone, m.p. 270°C, yield 60°₀. (Found: C, 53.98; H, 2.98; N, 21.20; C₂₄H₁₆N₈O₃S₂ requires C, 54.55; H, 3.03; N, 21.21 %); IR: 3410 (-NH-), 1700 (amide C=O), 1460 (-N=N-), 1380 (v_{sy} SO₂); PMR (δ, DMSO-d₆); 7.50-8.00 (12H, m, aromatic), 8.40 (s, 1H, -CONH-), and 9.00 (s, 2H, SO₂NH₂).

The characterization data of azoderivatives (IVa-f) are listed in table 1. All the compounds synthesised were screened for their antifungal activity against Curvularia lunata and Dreschlera halodes by following glass slide humid chamber method ¹⁰. Compounds III and IVc were effective against both the fungi at a dose level of 120 μ g/ml and 360 μ g/ml. Compounds IVa and

Table 1 Characterization Data of IV

Compound	R	mp(°C)	Y ield %	Molformula (Cryst from)
1V b	мн № —с-мн ₂	238	70	C ₂₅ H ₁₈ N ₁₀ O ₃ S ₂ (methanol)
t V c		230	62	C28H18N1003S2
IV d	OCH3 OCH3	280	78	C ₃₀ H ₂₂ N ₁₀ O ₃ S ₂ (AcOH)
/ ∀ •	CH3	260	72	C ₃₀ H ₂₂ N ₁₀ O ₃ S ₂ {AcOH}
144		>320	6.5	C ₃₃ H ₂₂ N ₁₀ O ₃ S ₂ (Aq alcohol)

^{*} All the compounds gave satisfactory analysis for C.H & N.

IVb have no fungicidal activity, while the remaining compounds were intermediate in their activity.

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- 1. Jeney, E. and Zsolnai, T., Zentr Bakteriol Parasitenk Abt I Orig, 1960, 177, 215; Chem. Abstr., 1960, 54, 21498h.
- Schonberg, A. and Singer, E., Chem. Ber., 1977, 110, 3945.
- 3. Ruhemann, S., J. Chem. Soc., 1910, 97, 1438.
- 4. Gagnaux, Pierre (Sandoz Ltd) Geroffen, 2,002, 892, 30 July 1970; Chem. Abstr., 1970, 73, 77285m.
- 5. Brown, P., Chem. Rev., 1961, 61, 463.
- 6. Rao, R. P., Indian J. Appl. Chem., 1960, 23, 110.
- 7. Sundholm, N. K. and Skapatason, J. B., US pat., 2, 510, 725, (June 6, 1950); Chem. Abstr., 1950, 44, 8045.
- 8. Kittleson, A. R., Yewck. H. L., Cohen, C. A., Hawley, R. S. and Smith, P. V., Brit. Pat., 716, 533 (Oct. 6, 1954); Chem. Abstr., 1955, 49, 12543.
- 9. Rajeswar Rao, V., Ashok Kumar, R. and Padmanabha Rao, T. V., Indian J. Chem., (in press).
- 10. Phytopathology (American Phytopathological Society, Committee on Standardization of Fungicidal Tests), 1947, 37, 354.

A SIMPLE COLORIMETRIC METHOD FOR DETERMINATION OF CARBOFURAN AND BENDIOCARB IN FORMULATIONS

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CARBOFURAN (Furadan, 2,3-dihydro-2,2-dimethyl-7-benzofuranyl methyl carbamate) and bendiocarb (1,3-benzodioxaol-40l, 2,2-dimethyl-methyl carbamate) are extensively used as insecticides and the former is highly toxic to mammals¹. Carbofuran is also used as a nematocide. The residues of these insecticides cause air and water pollution. Hence there is a demand for simple and rapid analytical procedure for the detection and determination of levels of residues in air and water environments. Colorimetric procedures for the determination of carbofuran in water have been reported²⁻⁵. In these methods carbofuran had been hydrolysed under alkaline conditions and the resulting