Structure of the Complex: The palladium chelate was analysed for its carbon, hydrogen, and nitrogen contents. The composition corresponds to the formula $Pd(C_9H_9O_2NCI)_2$ (Calc. C. 42.95%; H. 3.6%; N. 5.55%. Found C. 42.91%; H. 3.56%; N. 4.99%)

The normal oximes^{7,8} have ir absorption bands at 3300-3150 cm⁻¹ (OH Str.), 1690-1510 cm⁻¹ (C=N str.) and around 950 cm⁻¹ (N-O str.). In the present investigations, HMCAO has shown prominent bands at 3077 cm⁻¹, 1587 cm⁻¹ and 996 cm⁻¹ assignable to intramolecular hydrogen bonded (OH), C=N and N-O str. frequencies.

The broad weak band at 3077 cm⁻¹ due to intramolecular hydrogen bonded OH in the reagent is not observed in the complex. The band due to phenolic C-O vibrations at 1361 cm⁻¹ in the ligand is observed at 1389 cm⁻¹ in the complex. These observations suggest that ortho-hydroxy group of the base has taken part in the bond formation.

The C=N str. at 1587 cm⁻¹ in the free ligand is found at 1530 cm⁻¹ in the complex and can be interpreted as a consequence of coordination of nitrogen with the metal.

On the basis of above results structure (Fig. I) is assigned to the complex.

Fig. 1

Thanks are due to Dr. R. N. Gupta, Principal and Dr. B. C. Banerjee, Head of the Chemistry Department.

Chemistry Department, Hindu College, Moradabad, August 14, 1978.

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- 1. Beamish, F. E., Talanta, 1958, 1, 3; 1966, 13, 773.
- Halland, W. H. and Bozie, J., Anal. Chem., 1968, 40, 433.
- 3. Poddar, S. N., Anal. Chem. Acta, 1963, 28, 586.
- 4. Huston, R. H. and Chen, P. S., J. Am. Chem. Soc., 1933, 55, 4214.
- 5. Sulzberzer, N., Chem. Abst., 1924, 18, 3194.
- 6. Fries, K. and Flink Ber., 1908, 41, 4276.
- 7. Brown, I. F., J. Am. Chem. Soc., 1965, 77, 6431.
- 8. Golden, J. Chem. Soc., 1953, p. 997.

ISOQUINOLONES - AN ELEGANT SYNTHESIS FOR 3-ACETYL AND 3-BENZOYL-ISOQUINOLONES

Isoquinolones are growing in importance because of medicinal activity1-7 and also because of their potentiality as good intermediates for synthesis of isoquinoline alkaloids. We report easy synthesis for two such potentially useful compounds, viz, 3-acetyland 3-benzoy1-1(2H)-isoquinolones (IV and V) from homophthalic anhydride in two steps schematically below. 3-Acetyl-1(2H)-isoquinolone shown obtained12 before by action of ammonia on 3-acetylisocoumarin. The latter compound was obtained12 with difficulty by condensing potassium salt of phthalaldehydic acid with chloroacetone and cyclising the resulting acetonyl ester. The ease of oxidation of II and III by SeO₂ suggests the same mechanistic course as proposed13 before for oxidation of 3-methyl-isoquinoone to 3-formylisoguinolone.

3-Ethyl-1(2H)isoquinolone (II) is obtained in one step from homophthalic anhydride (I) (1 g) by stirring it in pyridine (3 ml) with propionyl chloride (1·3 ml) in chloroform (4 ml) which was added slowly drop by drop at 0°. The reaction mixture was stirred at 0° for 1·5 hr and then mixed with ammonia (25 ml) and heated on a boiling water bath for 2 hr and left aside at room temperature when 3-ethyl-1(2H)-isoquinolone (II) separated out as crystalline solid (0·2 g) [Crystallised from ethanol as colourless leaflets, m.p. $143-44^\circ$ Yield: 0·18 g. Found C, $76\cdot3$; H, $6\cdot7$; N, $8\cdot2$. $C_{11}H_{11}ON$ requires C, $76\cdot3$; H, $6\cdot3$ and N, $8\cdot1$. UV (MeOH) 224, 280, 330 nm; $\log \epsilon$: $4\cdot32$; $4\cdot06$; $3\cdot78$].

3-Ethyl-1(2H)isoquinolone (II) is obtained, however, in much better yield (0 5 g) by acylating homophthalic anhydride (1 g) with propionic anhydride (1·6 ml) and pyridine (1·2 ml), isolating the crude 4-propionyl-isochroman-1,3-dione (VI) (0·8 g) crude, m.p. 110-12°, needles from benzene-petrol ether, m.p. 113-15° (Lit.8 m.p. of pure 113-15°) and then heating the crude product with liquor ammonia (10 ml) on boiling water

bath for 3 hr. 3-Ethyl-1(2H)isoquinolone (II) (0.4 g) separates out from the reaction mixture in pure state which is filtered and additional amount (0.1g) is obtained by acidification of the filtrate. Synthesis of the compound VI by reglation is described before. but a modification of it improves the yield considerably. The modification consists of use of homophthalic anhydride in place of homophthalic acid, use of less amount of propionic anhydride and slightly different work up. The modified procedure is as follows: Homophthalic anhydride (I) is added slowly to a mixture of propionic anhydride and pyridine with stirring at room temperature and stirring is continued further for 2 hr. The compound VI that separates as yellow solid is filtered after stirring the reaction mixture with ether (14 ml).

The reaction that gave II in one step from I has thus proceeded through the intermediate VI and the conversion of VI to II takes place by the mechanism published earlier (The Tirodkar-Usgaonkar isoquinolone synthesis⁹). The compound II is also obtained in good yield by heating 3-ethylisocoumarin⁸ (VII) and 4-carboxy-3-ethylisocoumarin (VIII) with ammonia in boiling water bath as above. In case of VIII, replacement reaction of lactonic oxygen by NH is accompanied by decarboxylation (see ref. 9).

3-Ethyl-1(2H)isoquinolone (II, 1 g) is oxidised by refluxing it with selenium dioxide (1 g) in dioxane (40 ml) for 9 hr to furnish 3-acetyl-1(2H)isoquinclone (IV, 0.4 g) in good yield [colourless needles from ethanol, mp. 182-83°. Lit 12 m,p. same. Found C, 71.1; H, 5.1; N, 7.6. $C_{11}H_9$ O_2N requires C, 70.6; H, 4.8 and N, 7.5. UV (MeOH) 212, 250, 337 nm. $\log \varepsilon 4.22$, 4.0 and 4.01]. The product is isolated by filtering the reaction mixture hot from precipitated selenium and concentrating the solution when the compound crystallises out. 3-Benzoyl-1(2H) isoquinclone (V) is similarly synthesised by oxidation of 3-benzyl-1(2H)isoquinolone (III, 1g) by SeO₂ (1g) in dioxane (40 ml) for 4.5 hr in excellent yield [colourless needles from ethanol, m.p. 186-87°. Found C. 76.8; H, 4.4; N, 5.7. $C_{16}H_{11}O_2N$ requires C, 77.1; H, 4.4 and N, 5.6. UV 215, 257, 342; $\log \varepsilon$; 4.3, 4.04 and 4.05]. The constitution got confirmed from its IR and NMR Spectra. It gave IR bands at 1670 (C=O of ketone), 1650 (C - O of lactam), 1610, 1570 (aromatic) and NMR (CDCl₃) gave signals at $\delta 7.13$ (111, s, 11-4), 7.7 (8H, m, aromatic protons except H=8), 8.5 (1H, m, H=8), 9.8 (1H, NH). The aromatic proton II-8 resonated downfield due to C- O in proximity. The compound gave 2,4-DNP (yellow prisms) from ethanol, m.r. 283-84; Found N. 16.5; $C_{22}H_{13}O_{a}N_{a}$ requires N, 16.3) and an Oxime (needles) from benzene, m.p. 264-65°; Found C, 73·1; 11, 5.0; N, 10.8; $C_{1_0}H_{12}O_2N_2$ requires C, 72.7; H,

4.5 and N, 10.6). 3-Benzylisoquinolone (III) was synthesised before from homophthalic anhydride by one step method¹⁰ and by a two step method¹¹ in very good yields. The latter was via 3-benzylisocoumarin.

Microanalyses were carried out by Mrs. J. A. Patankar and Shri D. S. More. The I.R. spectra were recorded by Dr. P. M. Dhadke. The authors thank Dr. K. Nagarajan (Ciba-Geigy Research Centre) for NMR spectra and the University Grants Commission for financial assistance and research fellowship to ARM.

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- 1. Hamamoto, K. and Kajiwara, T., Japan Patent, 23,182 and 20, 558, 1965, Chem Abstr, 1966, 64, 3503e, 3505e.
- 2. Nakanishi, M. and Muro, T, Japan Patent, 7112,454 (Cl. Coyd, A 61K); Chem Abstr., 1971, 15, 35805.
- Sulkowski, T. S. and Wille, M. A., U.S., Patent 3,452,027 (Cl 260-289; CO7d); Chem. Abstr., 1969, 71, 112830t.
- 4 Tomizawa, H., Japan Patent, 71,39,701 (C107d A01n); Chem. Abstr., 1972, 76, 34123b.
- Buckle, D. R., Cantello, C. C. Barrie and Smith, H., Ger. Offen., 2,448,387 (Cl C07 D A61k), 1975; Brit. Appl., 47,485/73; Chem. Abstr., 1975, 83, 79080w.
- Coyne, W. E. and Cusic, J. W. (Searle, G. D. and Co.), U.S., Patent 3,600,394 (Cl 260-288 C07d); Chem. Abstr., 1971, 75, 118241f.
- Ueno, S., Sugihara, K., Hirayana, H. and Ichino, M., Japan Kokai, 76,101,985 (Cl C07D 217,08);
 Chem. Abstr., 1977, 86, 121169w.
- 8 Tirodkar, R. B. and Usgaonkar, R. N., Indian J. Chem., 1970, 8, 123.
- 9. and —, *Ibid.*, 1972, 10, 1060.
- 10. Modi, A. R. and Usgaonkar, R. N., Curr. Sci., 1976, 45, 832.
- Belgaonkar, V. H. and Usgaonkar, R. N., Indian.
 J. Chem., 1975, 13, 336.
- Kanevskaya, S. I., Kovsharova, I. N. and Linevich,
 L. I., Shornik Statel Obschehel Khim., 1953,
 2, 1493; Chem. Abstr., 1955, 49, 5478a.
- 13. Modi, A. R., Nadkarni, D. R. and Usgaonkar, R. N., Indian J. Chem. (In press).