Chemistry Department, Allahabad University, Allahabad 211 002, June 30, 1977. P. GANDHI.*
R. D. TIWARI.

- *Present address: P. Gandhi, Chemistry Department, The University of Texas at Austin, Austin, Texas-78712.
 - 1. Murti, V. V. S., Raman, P. V. and Seshadri, T. R., Tetrahedron, 1967, 23, 397.
 - 2. Nakazawa, K. and Ito, M., Chem. Pharm. Bull. (Tokyo), 1963, 11, 283.
 - 3. Jurd, L., Chem. and Ind., 1961, p. 322.
 - 4. Mathai, K. P. and Sethna, S., J. Indian Chem. Soc., 1964, 41, 347.
 - 5. —, Kanakalakshmi, B. and Sethna, S., Ibid., 1967, 44, 148.
 - 6. Kanakalakshmi, B., Ibid., 1969, 46, 279.
 - 7. Gandhi, P., Indian J. Chem., 1976, 14B, 1009.
 - 8. Ahmad, S. and Razaq, S., Tetrahedron Letters, 1971, 48, 4633.
 - 9. Gandhi, P. and Tiwari, R. D., J. Indian Chem. Soc., 1975, 52, 1111.
- 10. Molyneux, R. J., Waise, A. C. and Haddon, W. F., Tetrahedron, 1970, 26, 1409.
- 11. Gandhi, P. and Tiwari, R. D., Curr. Sci. 1976, 45, 18.
- 12. and —, Indian J. Chem., 1976, 14B, 632.
- 13. Nakazawa, K., Chem. Pharm. Bull. (Tokyo), 1962, 10, 1036.
- 14. Gandhi, P., D.Phil. Thesis, Allahabad University, 1974.
- 15. Natarajan, S., Murti, V. V. S. and Seshadri, T. R., Indian J. Chem., 1969, 7, 751.

SOME NEW HYDROXY SANDARACOPIMAR-15-ENES FROM PREMNA LATIFOLIA ROXB.

We report in this communication the isolation and constitution of four diterpene alcohols from the root-bark of Premna latifolia Roxb. (Verbenaceae). The residues obtained from n-hexane extracts were chromatographed over silica gel, after removal of latex, when the four compounds A, B, C and D separated out and their properties are recorded below. All these compounds gave red Liebermann-Burchard test.

Compound-A appeared as colourless needles from n-hexane, m.p. 42-43° ($C_{20}H_{34}O$; m/e M°, 290), $\{a\}_{11}^{10}$ -- 6·6° (e, 1·0, CHCl₃); no absorption in the U.V. above 200 nm; l.R. $v_{max}^{CCI_4}$ cm ¹ 3620 (- OH), 3090, 1820, 1640 and 910 (vinyl), 1390, 1370 (gem dimethyl) and ¹H NMR (XL-100 spectrum) CCl₄ (δ) 0·85 (s, 6H, two t-CH₃), 0·97 (s, 3H, t-CH₃), 1·19 (s, 3H, t-CH₃) and an ABX pattern characteristic of vinyl group (4·73, dd, J = 10·5, 1·5 Hz, 1H; 4·77,

dd, J = 17.5, 1.5 Hz, 1H and 5.63, dd, J = 17.5, 10.5 Hz, 1H). The presence of five terminal pendant groups, viz., four t-methyls and a vinyl group, besides its molecular formula suggest a pimarane, rosane or cassane type structure for compound-A. A survey of literature reveals that these characteristics are in agreement with those recorded for sandaracopimar-15-en-8 β -ol (I) which was isolated by Corbett et al. from Dacrydium colensoi and Bohlmann et al. 2.3 from Garuleum pinnatifidum DC., G. bipinnatum Less., Osteospermum fructicosum (L.) Norl., O. junceum Berg., O. corymbosum and O. rotundifolium.

Compound-B crystallised from n-hexane as colourless needles, m.p. $204-206^{\circ}$ ($C_{20}H_{32}O_2$; m/e M⁺, 304), $[a]_{D}^{30}$ -5.5° (c, 1.0, pyridine); no absorption in the U.V. above 200 nm; I.R. $v_{\text{max}}^{\text{KBr}}$ cm⁻¹ 3370 (- OH), 3(80, 1820, 1635 and 915 (vinyl), 1698 (six membered ketone) and 1375, 1395 (gem dimethyl) and ¹H NMR (XL-100 spectrum) CCl₄ (δ) 0.85 (s, 6H, two t-CH₃s), 1.22 (s, 3H, t-CH₃), 1.24 (s, 3H, t-CH₃), an ABX pattern of vinyl (4.86, dd, $J = 10.5, 1.5 \text{ Hz}^3$; 1H; 4.90, dd, J = 17.5, 1.5 Hz, 1H and 5.76, dd, J = 17.5, 10.5 Hz, IH) and 2.08-2.47 (broad, 4H). It did not form a 2, 4-dinitrophenylhydrazone, perhaps, the carbonyl is hindered. Further, it did not yield an acetate with Ac₂O/pyridine indicating that the hydroxyl is either hindered or tertiary. The presence of five terminal pendant groups, the molecular formula, besides the co-occurrence of I indicates that it could well be considered as sandaracopimar 15-en-8\beta-ol with a carbonyl function. The position of the carbonyl group could be established by dehydration using p-toluenesulphonic acid. when a liquid ($C_{20}H_{30}O$) was obtained, U.V. $\lambda \stackrel{EtOH}{=}$ 252 nm (log ε , 3.92) indicating α , β -unsaturated ketone further confirmed by I.R. $v_{max}^{liq, lilm}$ 1665 cm⁻¹ and absence of hydroxyl bands. It formed a 2, 4dinitrophenylhydrazone, m.p. 155-157° (C26H34O4N4). The ¹H NMR spectrum in CCl₄ showed no more olefinic protons besides the three vinylic hydrogens. thereby showing that the double bond formed was terrasubstituted and reasonably assumed to be $\Delta^{8,9}$. The carbonyl function could then be present in 7, 11 or 14 positions, and out of these II position was preferred in order to account for a single proton signal at 2.73 (8), dm, J. 13 Hz assignable "1.9 to CI-\$ II in this anhydro ketone, the constitution of which could be represented as sandaracopimar-8, 15 dien-11-one (11 b) and the starting keto alcohol is 8\beta hydroxysandaracopimar 15-en-11-one (II a). Sodium borohydride reduction of H a gave a diol (H c), m.p. 158 162 $(C_{810}H_{34}O_2)$. [a] 30 + 21.6° (e, 1.0, CHCl3) which neither reacted periodic acid nor could be acetylated with AciO! pyridine and showed in ¹H NMR spectrum a single proton signal at 4.44 d, broad w₁ 7 Hz assignable to methine proton of ~CHOH group, indicating that this hydrogen is equatorial and the hydroxyl β -axial. This diol is considered to be a CII-epimer of sandara-copimar~15~en-8 β , 11 α -diol, described by Bohlmann et al².

The compound-C crystallised as colourless needles from n-hexane, m.p. 160 ($C_{20}H_{34}O_2$, m_ie M¹, 306); $[a]_{D}^{30} - 15 \cdot 2^{\circ}$ (c. 2.5, CHCl₃); U.V. no absorption above 200 nm; I.R. y_{max}^{CC14} cm⁻¹ and 3480-3520 (hydroxyls), 3100, 1825, 1640 and 915 (vinyl) 1395, 1375 (gem dimethyl); ¹H NMR (XL-100 spectrum) CCl₄, δ , 0.85 (s, 6H, two t-CH₃^s), 0.99 (s, 3H, t-CH₃), 1.17 (s, 3H, t-CH₃), 2.34 (ddm, J = 14, 4 Hz, 1H, 3.24 (dd, J = 9.6 Hz, 1H) and anABX pattern of vinyl (4.72, dd, J = 10.5, 1.5 Hz,1H: 4.77, dd, J = 17.5, 1.5 Hz, 1H and 5.63, dd, J = 17.5, 10.5 Hz, IH). These data clearly suggest that compound-C also possesses sandaracopimar-15 $en-8\beta$ -ol moiety. Hydrogenation over palladium catalyst gave a dihydro derivative, m.p. 154.5° $C_{20}H_{36}O_2$). The second oxygen is present in sec. alcohol as a monoacetate. m.p. 108-111° (C₂₂H₃₆O₂, III b) and a ketone, m.p. $207-208^{\circ}$ (C₂₂H₃₂O₂, III c) could be prepared.

The methine proton of -CHOH in compound-C, which appeared at 3-24 δ was deshielded to 4.49 δ , t, J = 7 Hz in its acetate and the w_1 of 15 Hz of this signal in either case indicate that it is axially oriented. The compound neither cleaved with periodic acid nor formed an acetonide and further dehydration of its ketone did not yield an α , β -unsatured product. The splitting pattern of the methine signal, under these circumstances, suggests 1, 3 or 12 positions for the sec. hydroxyl. Sandaracopimar-15-en-8 β , 12 β diol, which was described earlier by Bohlmann et al.2 differs from compound-C. Secondary hydroxyl in 1-position was preferred in order to account for a single proton signal at 2.34δ which could well be assigned to 11a-hydrogen. The paramagnetic effect of 11α -OH on 1β -H in this series of compounds has been amply justified by Bohlmann and coworkers2,01,4

Now in the present case, 1β -OH exerts a similar effect on 11α equatorial proton. Further, sandaracopimar-15-en-8 β , 11β -diol (II c) has been described above and its 11α epimer was reported earlier² and these two compounds differ very much from compound-C which may now be represented as sandaracopimar-15-en-1 β , 8β -diol (III a).

The compound-D was obtained as colourless needles from benzene, m.p. $198 \cdot 5 - 200^{\circ}$ ($C_{20}H_{34}O_3$, m/e M⁺, 322); $[\alpha]_{13}^{30} = 33.6^{\circ}$ (c, 1.0, CHCl₃); U.V. no absorption above 200 nm; I.R. v_{max} cm⁻¹, 3480-3310 (hydroxyls), 3090, 1620, 910 (vinyl) and 1390, 1380 (gem dimethyl); ¹H NMR (XL-100 spectrum) CDCl₃, δ , 0.84 (s, 6H, 2 × t-CH₃), 1.04 (s, 3H, t-CH₃), 1·20 (s, 3H, t-CH₃), 2·34 (m, 1H), 3·39 (t, J = 7 Hz, 1H), 3.42 (t, J = 7 Hz, 1H) and an ABX characteristic of vinyl (4.81, dd, J = 10,pattern 1.5 Hz, 1H; 4.85, dd, J = 17.5, 1.5 Hz, 1H and 5.74, dd, J = 17.5, 10 Hz, 1H). On hydrogenation over Adams catalyst it gave the dihydrotriol, m.p. 208° (C₂₀H₃₆O₃); and on acetylation a diacetate (IV b), m.p. $165-166^{\circ}$ (C₂₄H₃₈O₅) in which the methine protons of -CHOAc appeared as triplets at 4.55 and 4.56 δ , J = 7 Hz indicating an axial configuration, besides these signals could be taken as X parts of ABX patterns. It could reasonably be assumed, in the light of the facts presented, that compound-D also belongs to sandaracopimar-15-en-8 β -ol series with two additional hydroxyls. From a study of the spectral characteristics of the derivatives and also a keto-diol, m.p. $180-182^{\circ}$ ($C_{20}H_{32}O_3$) obtained as a minor product in oxidation using pyridine-chromium trioxide, the two secondary hydroxyls could be tentatively placed at 1 and 12 positions (IV a); further investigation is in progress to establish this.

Fuller details will be published elsewhere.

The authors express their thanks to the CSIR, New Delhi, for the award of a Research Fellowship to T. N.

Department of Chemistry, Ch. Bheemasankara Rao. Andhra University, T. Namosiva Rao. Waltair 530 003, A.P., (India), January 5, 1978.

- 1. Corbett, R. E. and Smith, R. A. J., J. Chem. Soc. (C), 1967, p. 300.
- 2. Bohlmann, F., Weickgenannt, G. and Zdero, Ch., Chem. Ber., 1973, 106, 826.
- 3. and Zdero, Ch., Ibid., 1975, 108, 362.
- 4. Applications of NMR Spectroscopy, Illustrations from the Steroid Field by N. S. Bhacca and D. H. Williams (Holden-Day Inc.), 1964.