Potassium in shrink-swell soils of India

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This article reviews the information regarding potassium (K) status of Indian soils based on research conducted since 1929. The patterns and lacunae regarding potassium consumption in India are also mentioned. The role of minerals in potassium availability vis-à-vis forms of potassium is discussed and elucidated with suitable clay mineralogical evidences. The article also highlights the problems of potassium availability to plants in Indian shrink-swell soils. We have pointed out the inefficacy of the universal method used for assessing plant-available K (1N NH₄OAc) in Indian shrink-swell soils, as observed from extensive K response studies. The current practices of assessing only plant-available K is not adequate to detect native changes in soil potassium. This paradoxical situation necessitates revision and revalidation of the existing potassium fertilizer recommendations, which are being adopted since four decades. A holistic research envisaging soil test crop response and mineralogical studies will help in revising potassium evaluation methods in India, leading to judicious fertilizer application by the farmers.

Keywords: Clay mineralogy, fertilizers, mineralogy, potassium availability, swell–shrink soils.

THE shrink–swell (cracking clay) soils are primarily found in Peninsular India, expanding from 8°45′N to 26°0′N lat. and 68°0′E to 83°45′E long. Predominantly, these shrink–swell soils are developed from the weathering of Deccan lava¹ occupying an area of about 116 m ha in the country². These soils are smectite-rich with sufficient reserves to supply potassium (K) to the plants³.

Potassium is often considered as the quality element for crop production⁴, and is recognized as a quality-plus-yield nutrient. It helps promote photosynthates translocation as well as mobilization of stored material⁵. The major role of K is stomatal opening and closing which maintain the transpiration of water, and penetration of atmospheric CO₂ into the leaf which helps in improving water use efficiency⁶.

The net negative balance for NPK in the current agricultural scenario is 19% for N, 12% for P and 69% for K. The enormous proportion of potassium is because crops

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remove an average 1.5 times more potassium than nitrogen and application of K through fertilizer is considerably lower than that of N or P (refs 3, 7). The K balance is negative in almost all the intensive cropping systems in India, since the addition of K rarely matches its removal, resulting in huge K mining. This situation creates great pressure on non-exchangeable K for meeting the potassium requirements of the plants. The long-term intensive cropping systems, without K inputs, unfavourably affect K supply for plant uptake, ultimately hindering yield of the crops^{8,9}. Hence, it is inevitable to consistently focus on the role and importance of potassium in sustainable crop production by following balanced nutrient management. In order to maintain soil fertility, the removal of K needs to be balanced by sufficient K inputs. Hitherto, in India, there was a general consideration that black soils are rich in K and therefore K application was dispensable. However, with time it is predicted that in some soils deficiency of potassium could occur due to leaching loss, continuous cropping and soil erosion. In a nutshell, this article reviews the work done on potassium in Indian shrink-swell soils, helping to decipher the potassium behaviour from mineralogical signatures.

Potassium in soil

Potassium is an essential element for all types of plants and animals for completing their life cycle. It is the seventh most abundant element in the earth crust. The soils contain 0.04–3% potassium, the total K content of the upper 0.2 m of most agricultural soils generally ranges from 10 to 20 g kg⁻¹ (refs 9–11). However, most of the soil potassium (90–98%) is chemically bound in the crystal lattice structure of minerals, and thus not directly available or slowly available for plant uptake. The availability of K varies considerably from soil to soil, and is affected by different soil properties. Potassium in soil is categorized into four groups based on its availability to plants, viz. water-soluble, exchangeable, non-exchangeable/fixed and structural/mineral K.

Water-soluble potassium is the form of K which is directly available for plants and microorganisms; the K content mostly depends on depletion as well as replenishment of exchangeable and non-exchangeable

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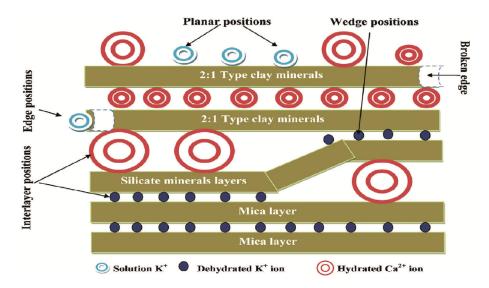


Figure 1. Potassium adsorption positions for K^+ in a mica-silicate mineral in the soil system (source: Meena *et al.*⁸⁶).

forms of potassium. The lower concentration of K in the soil solution induces the release of K^+ from non-exchangeable K (ref. 12). The K content in the soil solution ranges from 2 to 5 mg K I^{-1} for normal agricultural soils of humid regions I^{13} . The levels of solution K are affected by the kinetic and equilibrium reactions that occur between the K forms, the concentrations of bivalent cations on the exchanger and in solution phase and moisture content $I^{14,15}$.

Exchangeable K is the form of soil K which is electrostatically bound as an outer-sphere complex to the surface of clay minerals and humic substances¹⁶. The amount of K⁺ held by clay minerals at exchange sites depends on thermodynamic as well as kinetic factors¹⁷, and the affinity of the exchange sites for K⁺ is related to the concentration and nature of the soil surface of K⁺ in relation to the other exchangeable and mostly divalent cations present in the soils¹⁸. It is exchanged with other cations and is also readily available to plants¹⁹.

Non-exchangeable/fixed K is held between the layers of clay minerals and is not easily available for exchange with other cations. It is held between adjacent tetrahedral layers of dioctahedral and trioctahedral micas, vermiculites and intergrade clay minerals such as chloritized vermiculite 11,14,20. The binding forces between K and the clay surfaces are greater than the hydration forces between individual K+ ions, resulting in potassium fixation. This results in a partial collapse of the crystal structures and the K⁺ ions are physically trapped to varying degrees, making K release a slow, diffusion-controlled process¹¹. Several types of adsorption sites for K⁺ on clay minerals have been postulated (Figure 1)²¹. Those on planar surfaces (p-position) have low K⁺ selectivity; however, the edge (e-position) and wedge (w-position) have medium K⁺ selectivity. The K⁺ present in interlayer edge and wedge positions (Figure 1) is known as fixed K. The factors responsible for K fixation are the particle size distribution, types and quantities of clay minerals, and the addition or removal of K from minerals²².

Structural K is also known as mineral K. It is bonded covalently within the crystal structure of K-bearing minerals like micas and feldspars²³. Structural K is generally assumed to be only slowly available to plants; however, the availability is dependent on the level of K in the other forms, and the degree of weathering of the micas and feldspars constituting the mineral K fraction^{11,14}. The K forms of the soil in the order of their availability to plants are: solution > exchangeable > non-exchangeable (fixed) > structural^{11,14,15}.

Status of potassium in Indian soil

Potassium is one of the three main nutrients of balanced fertilizer use, coupled with nitrogen and phosphorus. India is the third largest user of N, P and K fertilizers in the world, with an annual consumption at about 18 million tonnes (mt) of $N + P_2O_5 + K_2O_7$, of which K comprises only one-seventh of the total^{3,24}. The net negative balance for N, P and K in the current agricultural scenario is 19% N, 12% P and 69% K. This huge negative difference of potassium is partly because crops remove an average of 1.5 times more potassium than nitrogen, and the application of potassium through fertilizers is considerably lower than that of N or P (refs 3, 7). The assessment of potassium status of Indian soils in 1929 suggested that soil K status was sufficient. Stewart²⁵ partially modified this and reported that most of the soils have adequate potassium, except the lateritic soils. Tamhane and Subbiah²⁶ reported that, out of 35,000 soil

Table 1. Status of potassium (K) in shrink-swell soils (Vertisols and Vertic intergrades) of India

		•	ean within brackets) ^e kg ⁻¹)			
Sampling location ^a	Soil type	Water-soluble K ^b	Exchangeable K ^c	HNO ₃ K	Non-exchangeable K^d	Source (ref.)
Nabibagh, Bhopal	Typic Haplustert	5–13	171-252 H	_	Н	87
Raipur, Chhattisgarh	Vertisols	16.3	102 M	_	788 H	88
Jhalawar, Rajasthan	Vertisols	98	121 H	743	637 H	89
Rajkot, Rajasthan	Vertisols	1–4	21-157(81) L-H	182	90-319 (182)	36
Indore, Madhya Pradesh	Vertisols	4-16(10)	101-147(132) M-H	_	_	36
Rewa, Madhya Pradesh	Vertisols	6-14 (9)	141-208(171) H	_	_	90
Vidharbha, Maharashtra	Vertisols	8	135 H	_	>600 H	91
Akola, Maharashtra	Vertisols	1-2(1)	13-69 (32) L-M	_	M	36
Solapur, Maharashtra	Vertisols	2-5	198-256 (217) H	520	640-720 (661)	36
Andhra Pradesh	Vertisols	2-4	Н	H	>600H	92
Telangana and Rayalseema districts	Vertisols	15	Н	-	Н	93
Guntur, Andhra Pradesh	Vertisols	6	94 (M)	_	Н	94
Karnataka (north)	Vertisols	20	212 (H)	_	625 (H)	95
Sadhugarh, Punjab	Vertic Ustochrepts	_	221.2 (H)	_	Н	96
Sehore, Madhya Pradesh	Vertisols	7.7	Н	821	H	88
Nimone, Maharashtra	Udic Chromusterts	_	310 (H)	_	540 (M)	97
Dhule, Maharashtra	Typic Chromusterts	12.4	455 (H)	1072	617 (H)	98
Western Gujarat	Black soil	5.98	148 (H)	499	383 (M)	99
Saurastra, Gujarat	Black soils	11	171 (H)	610	438 (M)	100
Gujarat (Pithvajal, Amreli)	Vertic Ustochrept	13	H	580	600-800	42
Kamliakheri, Indore, Madhya Pradesh	Vertic Ustochrept	9	265 (H)	800	>1000(H)	65

aNumber of states or agro-ecological regions/locations sampled. bMean of water-soluble K. cValues within brackets represent mean of exchangeable K (available K water – soluble K). dNon-exchangeable K is HNO_3 K minus exchangeable K. eIn some cases, range is given with or without mean. Values within brackets represent the mean. Exchangeable K: 0–50 mg kg⁻¹ low (L), 50–110 mg kg⁻¹ medium (M) and above 110 mg kg⁻¹ high (H). Non-exchangeable K: 0–300 mg kg⁻¹ low (L), 300–600 mg kg⁻¹ medium (M) and above 600 mg kg⁻¹ high (H).

samples analysed from different states in India, 64% were low to medium in potassium. Thereafter Ramamoorthy and Bajaj²⁷ reported that out of 1.3 million soil samples analysed, 20% were low, 53% medium and 27% were high in K status. The available potassium status generated in 1976 provided the benchmark information²⁸. Most of the Indian soils are categorized as medium to high in available potassium status. Motsara and Singh²⁹, and Motsara³⁰ have also assessed the K fertility status of Indian soils. The nutrient index values based on more than 11 million soil test data from 371 districts, showed that 76 (21%) were low in K, 190 (51%) were medium and 105 (28%) were high in K (ref. 31). Muralidharudu et al. 32 studied K fertility status of 500 districts and reported that 9% districts were low in K, 42% were medium and 49% were high in K. These data are based on 1N NH₄OAc method; the soils containing 130 kg K₂O/ha were categorized as low, 130-335 kg K₂O/ha as medium and above 335 kg K₂O/ha were categorized as high³³. Shrink-swell soils comprising Chromusterts, Pellusterts and Vertic Ustochrepts are found in the central highlands (Malwa, Budelkhand and East Satpura), Gujarat plains, Kathiawar peninsula, Deccan and the Eastern Ghats. They are high in exchangeable and medium to high in non-exchangeable K (ref. 34). To categorize the shrink-swell soils in different regions on the basis of available K (1N NH₄OAc extractable as described by Jackson¹⁰), non-exchangeable K (1N boiling HNO₃ extractable minus 1N NH₄OAc extractable as described by Jackson)¹⁰, water-soluble K (determined after extracting K from soil in water and using a flame photometer) and exchangeable K (derived by subtracting water-soluble K from available K; Table 1), the following norms were employed:

- Exchangeable K (mg kg⁻¹): low (0–50), medium (50–110) and high (>110)³³.
- Non-exchangeable K (mg kg⁻¹): low (0–300), medium (300–600) and high (>600)^{35–37}.

Pattern of potassium consumption and nutrient mining in India

The annual consumption of K was insufficient around 3000–6000 tonnes in the early 1950s and it reached up to 0.3 mt after 20 years. The potassium consumption increased from 0.3 to 1.36 mt between 1971–72 and 1991–92 (ref. 38). Temporal data for more than 50 years in India showed that potassium contributes to less than 10% of the total fertilizer nutrient consumption³⁹. The consumption of K is only 22% when compared to nitrogen consumption.

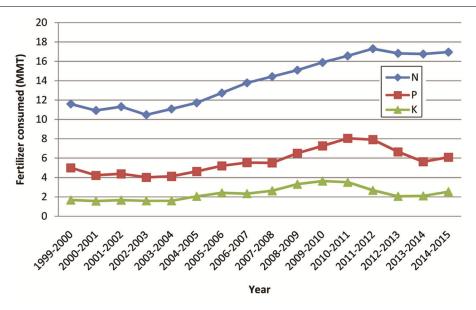


Figure 2. Trend of fertilizer consumption (N, P and K) in India.

The potassium consumption dropped from 1.36 mt in 1991-92 to 0.88 mt in 1992-93 due to decontrol, which increased the price of potassium fertilizers. After realizing the problem, the Government of India (GoI) had introduced a concession scheme to increase potash consumption by lowering maximum retail prices. Thereafter, potassium consumption increased and touched a record high of 3.63 mt in 2009-10 (Figure 2), and by 2025 K₂O consumption may be doubled to meet the foodgrain needs of the country. India has no deposits of K fertilizers, and all of them have to be imported. However, the major concern is the rapidly increasing price of muriate of potassium (MOP). It was Rs 5055/mt in 2010-11 and rose to Rs 17,892/mt in 2014-15 and continues to rise (Table 2)⁴⁰. According to the latest report of the Cabinet Committee on Economic Affairs, with the recent revision in nutrient-based subsidy (NBS) rates, the MRP for MOP could increase by about Rs 800-1000/mt.

Currently, GoI is spending about Rs. 9280/tonnes on K fertilizers to supply the farmers (Table 3)⁴¹. The consumption of N and P registered a decline of 3.7% and 3.9% respectively, during 2016-17 over the previous year. However, potassium consumption increased by 4.4% during 2016-17 (ref. 41). This scenario underlies the need for using the costly K fertilizer judiciously and most economically, considering the crop K needs and soil K reserves⁴². In case of nutrient mining, the balance sheets of nutrients in most of the Indian soils have been negative, and continue to be so. Based on nutrient balance sheets data published in Fertilizer News, Tandon' has condensed the state-level nutrient balance sheets (Table 4) and soil nutrient balance sheets of India (Table 5). The balance is positive for nitrogen and phosphorus, whereas it is negative for potassium on a gross basis. The net figures arrived after adjusting fertilizer input use efficiencies was 50% for N, 35% for P_2O_5 and 70% for K_2O (Table 5). In case of removal, 60% crop uptake was for K only, whereas for both N and P it was 80% (Table 5). As evident from Table 5, potassium has maximum negative balance followed by nitrogen, whereas it is lowest for phosphorus.

Clay mineralogy vis-á-vis forms of potassium

The potassium release from clay minerals is influenced by chemical composition and particle size of these minerals⁴³. It is a well-accepted fact that trioctahedral micas, such as phlogopite and biotite, release potassium more easily than dioctahedral micas, such as muscovite⁴⁴. Several studies have demonstrated the relationship between clay mineralogy composition and forms of potassium^{45–50}. The knowledge of clay mineralogy is essential for understanding the elemental composition and nutrient supplying power of the soils. K dynamics is principally influenced by the mineralogical composition of soils⁴⁸. The relationship between clay mineralogy and potassium forms can be used in evaluating potential soil K fertility, prediction of clay cycling and plant uptake⁴⁵. Information on exchangeable K (1N NH₄OAc) combined with knowledge of clay mineralogical composition can provide an understanding about the equilibrium and release of non-exchangeable K to plants and the need for potassic fertilizers⁴⁶.

Soils abundant in 2:1 clay minerals such as high-charge smectites, vermiculite and micas contain a huge amount of HNO₃-extractable potassium than kaolinite and other siliceous minerals^{45,47,48}. Bhonsle *et al.*⁴⁶ reported that kaolinitic soils had low levels of exchangeable K (1N NH₄OAc), illitic and mixed soils had medium level, and

Table 2. Retail price (MRP) in Rs/MT of Muriate of potassium (MOP) under the NBS regime

	Price (quarter-wise, Rs/million tonnes (MT)					
Year	I	II	III	IV		
2010-11	5055	5055	5055	5055		
2011-12	6064	11,300	12,040	12,040		
2012-13	16,695	23,100	24,000	18,750		
2013-14	18,638	17,750	17,750	17,750		
2014-15	17,892	17,892	17,892	16,980		

Source: Ref. 40.

Table 3. Subsidy on MOP fertilizer during 2011–12 to 2015–16

Year	Subsidy (Rs/tonne)
2011–12 2012–13 2013–14 2014–15 2015–16	16,054 14,400 11,300 9,300 9,300
2016-17	9,280

Source: Ref. 41.

smectitic soils had high levels of exchangeable K. Sharp-ly⁴⁵ reported that determination of both non-exchangeable and exchangeable K could give better indication of potential potassium supplying power of the soil.

Role of minerals in potassium availability

The primary K-bearing minerals in soils are micas and K-feldspars. Micas are more important than potassium feldspars in supplying potassium to plant^{20,51}. Plantavailable potassium, however, is related to the weathering of micas and feldspars in soil environment⁵¹. The contribution of potassium to the soil pool from feldspars is meagre because the process of K release is a very slow reaction in natural soil environment^{52,53}. In micas, a large unhydrated K-ion is held in ditrigonal cavities of the basal plane of oxygen of the tetrahedral sheets by electrostatic bonds. This keeps the layer collapsed⁵¹. Trioctahedral micas release their K by weathering more easily than dioctahedral micas⁵⁴. Serratosa and Bradly⁵⁵ observed that the proton of OH in trioctahedral micas is repelled equally by all the three octahedral cations, and lies on the normal to the basal plane, directed towards the interlayer space. In dioctahedral micas the proton is attracted to the vacant octahedral site and is displaced to one side of the normal. K in trioctahedral micas is repelled to a certain extent by the proton, and is therefore in a less electronegative environment than K in dioctahedral micas⁵⁶. This implies that K is held more strongly in dioctahedral mica as compared to trioctahedral micas.

The concentration of potassium in soil solution is the most important factor that controls the rate of release of

K from micas⁵¹. Pal et al.⁵⁷ reported that performing experiments to relate K release reaction of Indian soils without ascertaining the nature and composition of soil micas may not be adequate to establish a relationship between crop response to potassium fertilizers and nature of soil micas. Mica, hydrous mica and vermiculite have high adsorption/fixation properties^{16,58-60}, whereas smectites do not fix or adsorb K due to their low charge^{59,61-63}. Research on the fundamental aspects of potassium release and adsorption/fixation in relation to the mineralogy of soils gained momentum during 1985 to 2001 in India^{64,65}. The role of potassium-bearing minerals in releasing K from their non-exchangeable fraction of Indian soils is well established 42,66-69. Micas are prime K-bearing minerals of major Indian soils that are mainly concentrated in their clay and silt fractions^{59,70–74}. Despite this favourable K-mineral inheritance, crop response to potassium fertilizers in many of these soils has been anomalous^{59,73–75} Both trioctahedral and dioctahedral micas are common in these soils^{59,73-75}. Therefore, release of K from finegrained micas is not similar because they are far from ideal in composition and structure⁵¹. The proposed relationships between potassium release and micas are based on results obtained from specimen micas and not from soil micas, and thus they are speculative 20,60,76,77. The mica present in black soils contains both biotite and muscovite minerals⁷⁴, which is confirmed from the ratio of peak heights of 001 and 002 mica basal reflections. A ratio greater than one suggests the muscovitic characteristic of mica, but in reality, it indicates the presence of both biotite and muscovite minerals⁷⁴. The ratio would have been very close to unity if muscovite minerals alone are present⁷⁸. Pal et al.⁶⁰ observed a significant positive correlation between cumulative K release from sand, silt and clay, and their total K content, indicating that potassium release is a function of total K content in feldspars and micas. He further observed a positive correlation between total potassium content in soil, sand, silt and clay, and also highlighted the influence of mica to supply potassium to the plants grown in black soils. However, better correlation between potassium release in soil, sand, silt and clay and their biotite content provided undeniable evidence that the K release in soils is primarily controlled by biotite⁶⁰.

Table 4. Nitrogen (N) and potassium oxide (K2O) addition, removal and balance in major states ('000 t)

	N			K_2O			
State	Addition	Removal	Balance	Addition	Removal	Balance	
Maharashtra	923	1559	-636	197	2096	-1899	
Uttar Pradesh	2387	1497	889	114	1777	-1664	
Rajasthan	547	835	-288	7	1068	-1061	
Madhya Pradesh	519	696	-177	24	849	-825	
Punjab	1081	589	492	19	764	-745	
Andhra Pradesh	1256	477	779	191	817	-625	
Haryana	597	362	235	5	490	-485	
Karnataka	681	473	209	216	604	-388	
Gujarat	691	340	351	61	426	-365	
Odisha	196	227	-31	40	282	-242	
Tamil Nadu	484	405	79	162	398	-236	
West Bengal	562	764	-202	226	801	-575	
All India	10,923	9613	1310	1454	11,657	-10,203	

Source: Tandon⁷.

Table 5. N, P and K balance sheets in India

	Gros	s balance sheet ((000 t)	Net balance sheet ('000 t)		
Nutrient	Addition	Removal	Balance	Addition	Removal	Balance
N	10,923	9613	1310	5461	7690	-2229
P_2O_5	4188	3702	486	1466	2961	-1496
K_2O	1454	11,657	-10,202	1018	6994	-5976
NPK total	16,565	24,971	-8406	7945	17,645	-9701

Source: Tandon7.

Status of potassic fertilizer consumption in India

It is well known that K consumption in India mainly depends on availability of potassium, maximum retail prices and availability of potassium-carrying complexes in the region³⁸. During the last one and a half decades, urea availability has been satisfactory; but it was unsatisfactory in case of other fertilizer products. During 2016-17, import of urea and diammonium phosphate (DAP) reduced significantly from the level of the previous year. Import of MOP, however, showed an increase during the period. Import of urea was 5.48 million metric tonnes (MMT) in 2016–17 as against 8.47 MMT in the previous year. Import of DAP also declined to 4.39 MMT in 2016-17 from 6.01 MMT in the previous year. However, import of MOP increased to 3.74 MMT in 2016-17 from 3.24 MMT in the previous year⁴⁰. As we know Indian companies import potassium from Jordan, Canada, Belarus, the UK, CIS, Germany and Israel. Total imports of MOP in 2007-08 were 44.21 lakh tonne, which increased to 56.7 lakh tonnes in 2008-09. Table 6 shows Indian import statistics of potassium fertilizers from various countries.

Table 7 shows the quantity of nitrogen, phosphorus and potassium fertilizers imported during the last 12 years (up to January 2015). Fertilizers like DAP, urea, SSP and various grades of complex fertilizers are being produced

in the country. However, the gap between estimated requirement and indigenous production is being met through imports. MOP is the only fertilizer whose requirement is fully met through imports, as there are no feasible existing sources of MOP in the country⁴⁰. Scarcity in the availability of K and K-carrying complexes is the major reason for import and skewed NPK ratio in India. In fact, historically, nutrient management in India has been N-driven, followed by P and very less of K, which is the key reason for diminishing low nutrient use efficiency⁷⁹. The low use of potassium in crops is due to the perception that the Indian soils are rich in K, and reduction in potassium use will help the national exchequer as India imports its entire potassium fertilizer requirement^{80,81}. This has resulted in neglected use of K fertilizers by farmers⁸².

It is worth noting that even the most productive and progressive states like Haryana and Punjab have skewed N:P:K ratios. Traditionally, the focus has been on nitrogen application followed by phosphorus with very less use of potassium, resulting in a huge imbalance. The deficiency of potassium in Indian agriculture in 2020 is predicted to be around 10 mt/annum, whereas estimates of nitrogen and phosphorus balances are positive^{65,67}. The general tendency of Indian farmers is to purchase whichever fertilizer is available in the market. The effect of N and P fertilizers is seen within a short time of

Table 6. Import of K fertilizer in India

Quantity (in lakh metric tonnes)							
Country	2003–04	2004–05	2005-06	2006-07	2007–08	2008-09	Percentage imports in 2008–09
CIS*	10.98	15.36	23.57	16.24	12.58	21.09	37
Canada	4.51	6.45	8.12	7.02	10.66	14.04	25
Israel	2.55	6.4	6.89	4.82	11.25	11.43	20
Jordan	6.58	4.79	5.5	4.17	7.66	7.45	13
Germany	1.18	0.91	1.43	1.26	1.65	1.53	3
UK	_	_	_	_	0.19	1.12	2
Total	25.8	34.09	45.78	34.48	44.21	56.72	

^{*}Commonwealth of independent states. Source: Anon. 101.

Table 7. Import of N, P and K fertilizers in India

	Quantity (in lakh metric tonnes)						
Year	Nitrogenous fertilizer	Phosphatic fertilizer	Potassic fertilizer				
2003-04	132	338	1548				
2004-05	411	296	2045				
2005-06	1385	1122	2747				
2006-07	2689	1322	2069				
2007-08	3707	1391	2653				
2008-09	3751	3067	3403				
2009-10	3447	2756	2945				
2010-11	4493	3802	4069				
2011-12	5240	4427	3335				
2012-13	3505	2625	1178				
2013-14	3920	1588	1926				
2014-15	4766	1832	2537				

Source: Department of Fertilizers, Government of India⁴⁰.

application, whereas the effect of K application is seen only at maturity stage in the form of improvement in quality, shape, size and colour. Therefore, Indian farmers neglect potassium application in their fields. Hence, there is an urgent need to assess the expected responses to applied potassium so that potassium fertilizer management can be taken up for the efficient use of K and the resulting economy in potassium use⁶⁵.

Need for demand-driven research on potassium in shrink-swell soils

Anamolies exist in shrink–swell soils as observed from the inconsistent crop response to applied K fertilizers. Generally the soils which contain less than 120 kg ha⁻¹ K are categorized as low in available potassium, between 120 and 280 kg ha⁻¹ as medium and above 280 kg ha⁻¹ as high in available K (refs 3, 33). These rating limits are ambigious as they are classified irrespective of soils or crops. Solankey *et al.*⁸³ studied the response of two wheat varieties to K on farmer's field in black soils and reported that though these soils were sufficient in NH₄OAc–K, the crop responded to 30 kg ha⁻¹ K₂O. They have determined a critical limit for water-soluble K (14.4 kg ha⁻¹), but

failed to determine a critical limit based on NH₄OAc-K. Raheb and Heidari⁸⁴ observed that NH₄OAc–K extraction method does not give uniform trends in all mineralogical suits; a combination of both NH₄OAc-K and HNO₃-K gives a more applicable index for illitic, vermiculitic and smectitic soils. This indicates that the NH₄OAc method does not give proper indication of the viability of K in shrink-swell soils. Singh and Wanjari85 reported that soybean started showing a response to applied potassium fertilizer when available potassium status reached 316 kg ha⁻¹. However, wheat showed a response to potassium even earlier. Likewise, sorghum started showing response to applied potassium at 324.4 kg ha⁻¹, while a huge response in both the crops was noted at available potassium status of 307.6 kg ha⁻¹ in Akola, Maharashtra. Considering this crop response variation, we could take critical value for potassium as 312 kg ha⁻¹ (average of 307.6 and 316 kg ha⁻¹), but this value is greater than the threshold value 280 kg ha⁻¹ which is used to categorize the soil as high in available potassium. These results showed that in some situations, crop response to applied K may be poor in low K-containing soils, whereas significant responses to applied K are recorded in high Kcontent soils. These anomalies need to be addressed for judicious K-fertilizer management.

The critical levels to categorize soils into low, medium and high levels were set 50 years ago based on soil tests in the Indo-Gangetic plains. For example, Vertisols of central and southern India are categorized under high potassium level which leads to low potassium recommendation in this region. However, Vertisols with high clay content require more K compared to light textured soils of North India. Therefore, the yields are affected due to inaccurate recommendation for potassium application in Vertisols. Considering these prevailing anomalies, it has become need of the hour to revise the methodology for available potassium (1N NH₄OAc) in the soil, which is routinely followed throughout the country. Besides, the non-exchangeable K, mineralogical composition, Kfixing capacity as influenced by quality and quantity of clay under different soil, crop, and climatic conditions should also be taken into consideration for potassium recommendation. Therefore, the novel, holistic approach should envisage soil test crop response studies along with revision of existing K-testing methods, particularly for shrink-swell soils.

Conclusion

The available potassium status showed a gradual decline from medium to low in Indian shrink-swell soils. The routinely followed ammonium acetate method for potassium availability causes misinterpretation in shrink-swell soils of India. The fallacy that black soils (Vertisols) are rich in potassium has led to imbalanced potassium fertilizer recommendation causing deficiency in plants and soils. This paradoxical situation necessitates revision and revalidation of the existing potassium fertilizer recommendations, which are being adopted since four decades. India imports the entire quantity of potassic fertilizers, which gives an alarming call for judicious fertilizer management. A demand driven holistic research envisaging soil test crop response and mineralogical studies is the need of the hour for revising potassium evaluation methods in India. This research approach will optimize the input cost of K fertilizers and maintain soil health through judicious fertilizer management which will benefit the farmers in the long run.

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